

Bentonite microbial activity in geological disposal of radioactive waste

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**Nuclear Waste
Services**



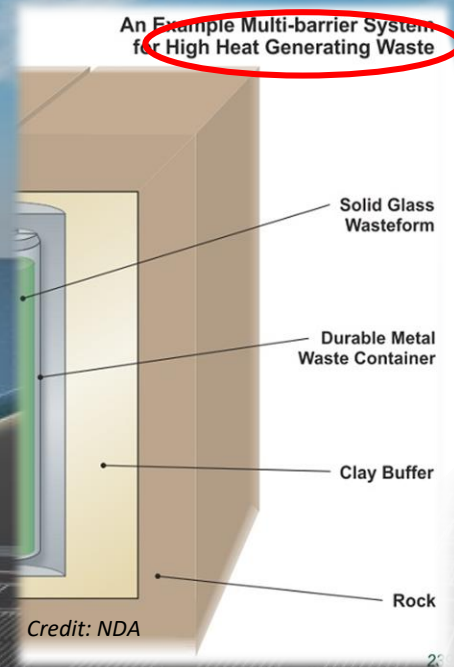
NNUF RADER

A facility supporting research into
Radioactive waste **D**isposal
and **E**nvironmental **R**emediation



**Research
Support
Office**

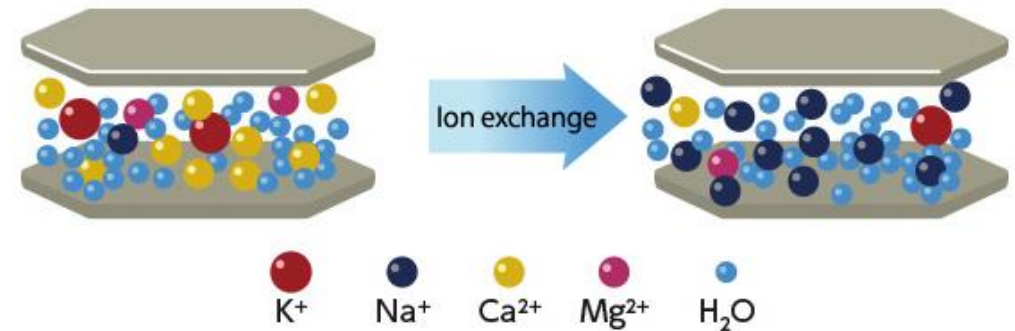
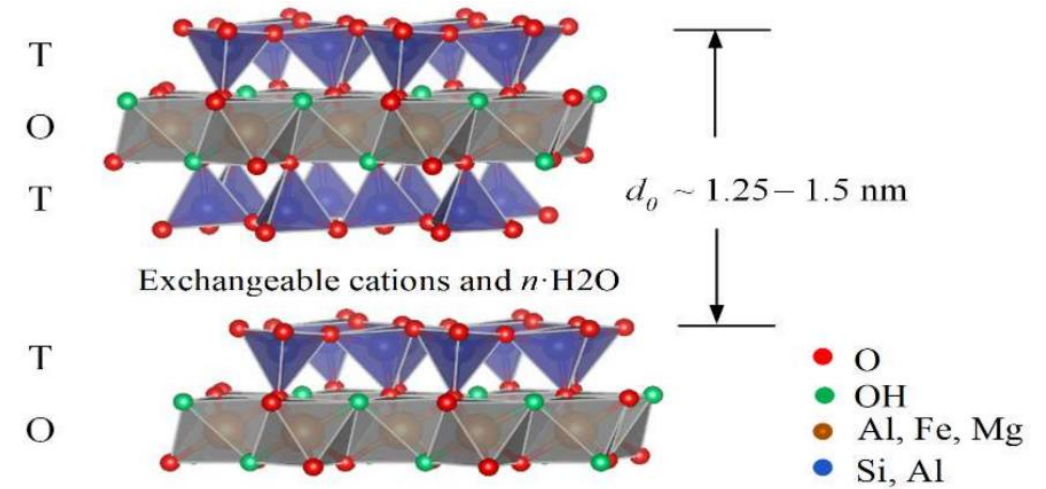
Geological Disposal Facility- GDF



Near field systems: canister/ bentonite interface

Bentonite – overview

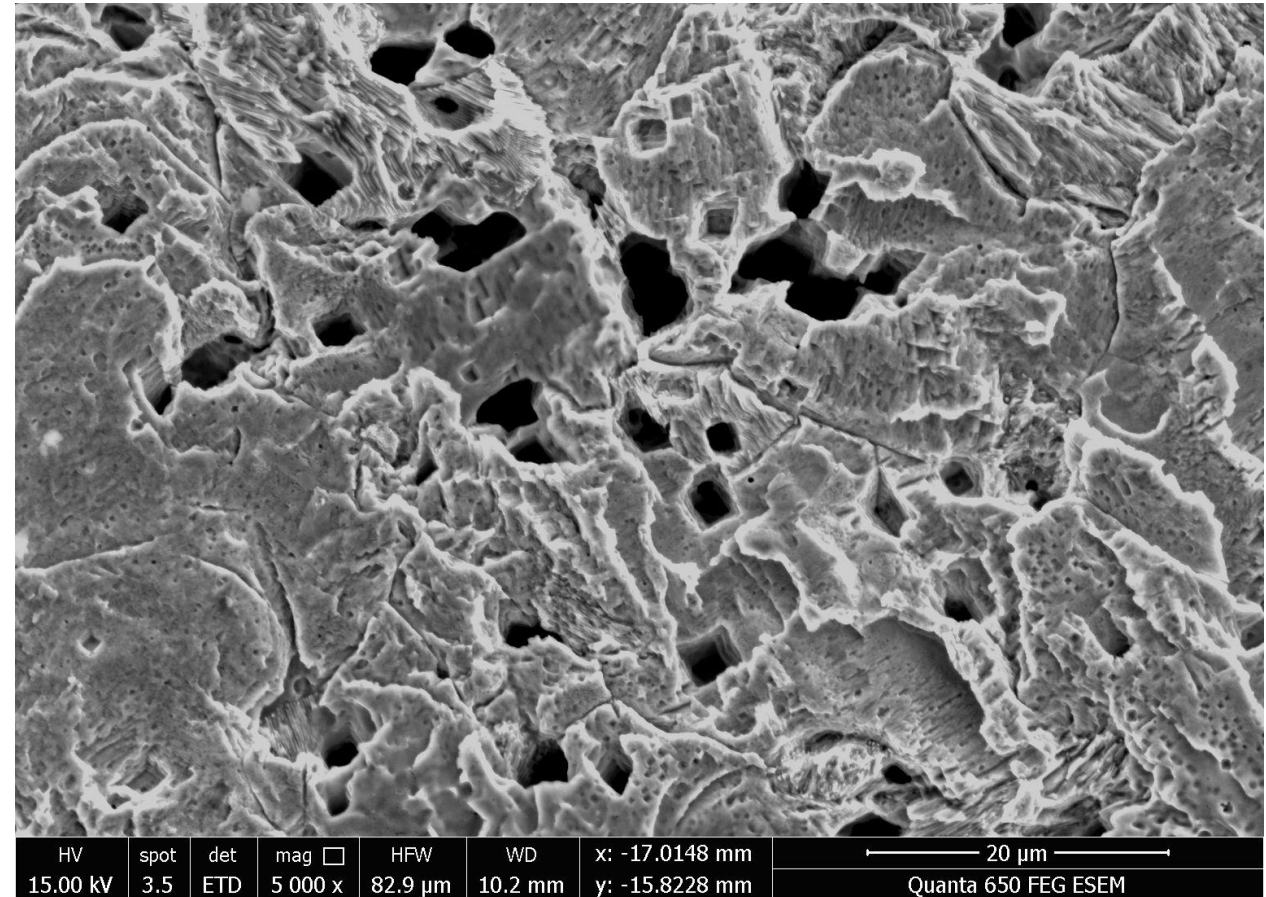
- Low permeability
- High sorption capacity
- (mostly) MONTMORILLONITE
- Swelling capacity
Due to the presence of the smectite mineral, montmorillonite
- Buffering capacity



Schematic diagram of cation exchange of montmorillonite

Bentonite microbial processes

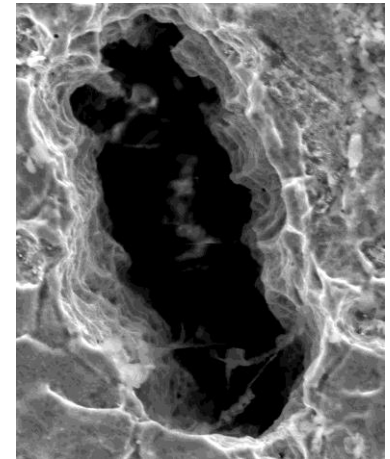
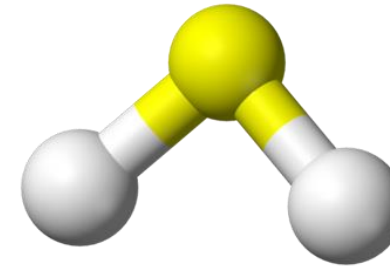
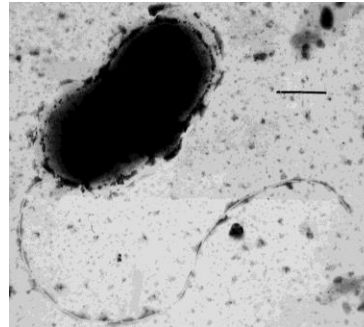
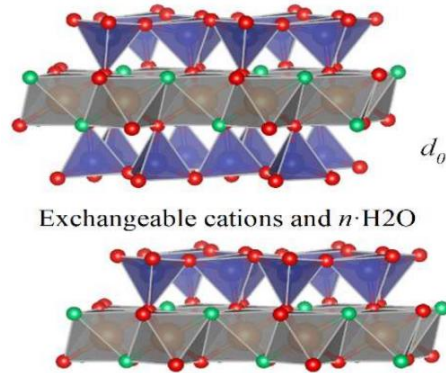
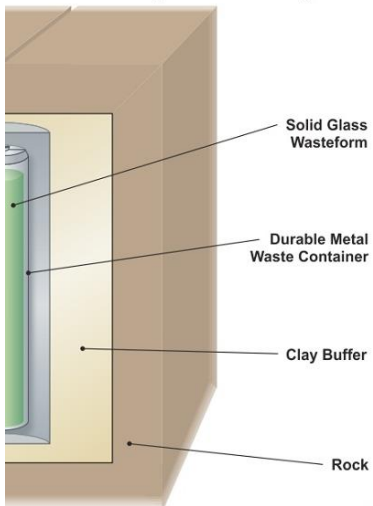
- Bentonites naturally contain **sulfate-reducing bacteria (SRB)**
- SRB require an **electron donor** and an **electron acceptor**
- **Sulfate (electron acceptor)** is present in groundwater
- Anoxic steel corrosion produces **hydrogen (electron donor)**



Microbially influenced corrosion (MIC)



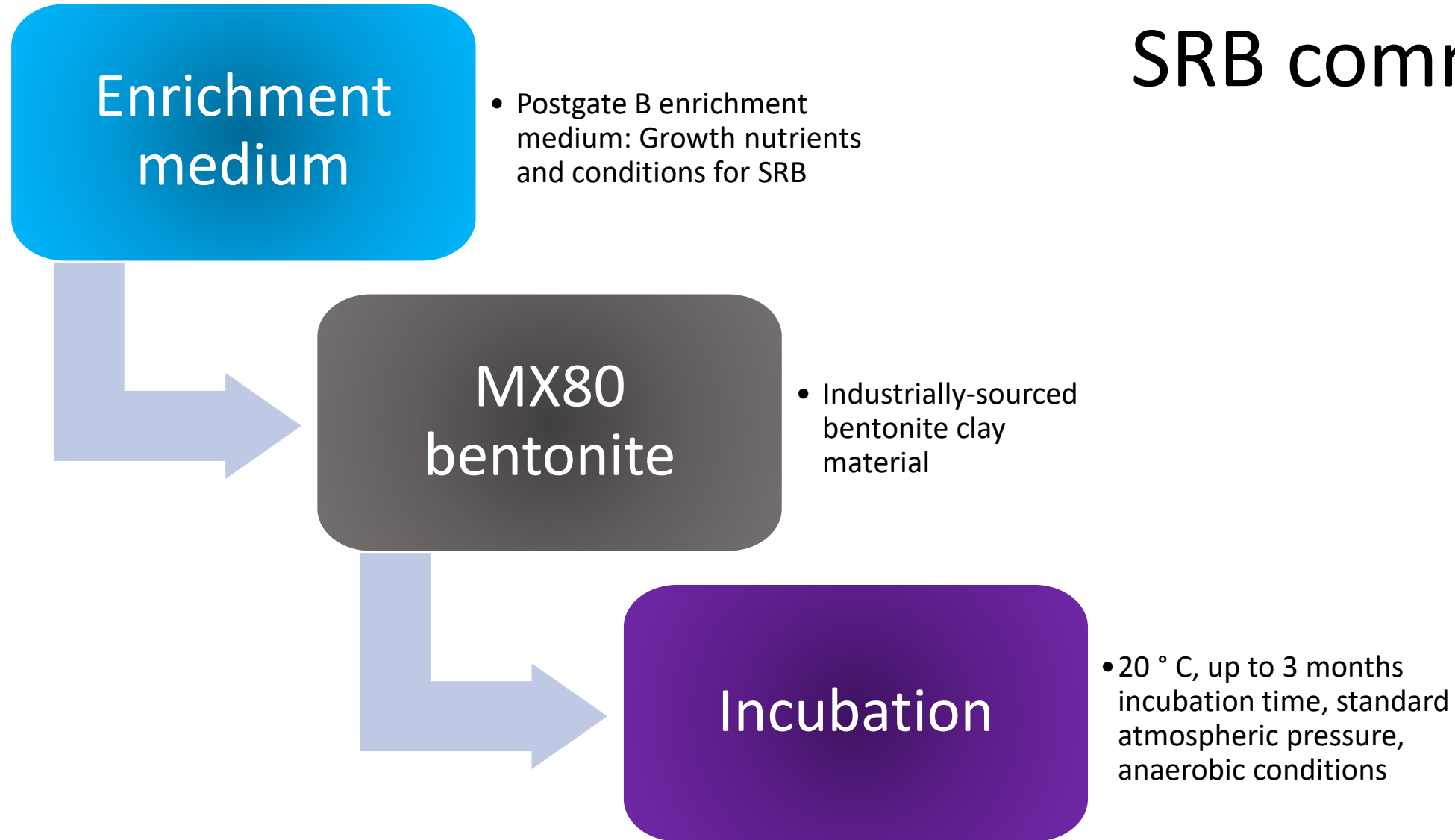
An Example Multi-barrier System for High Heat Generating Waste



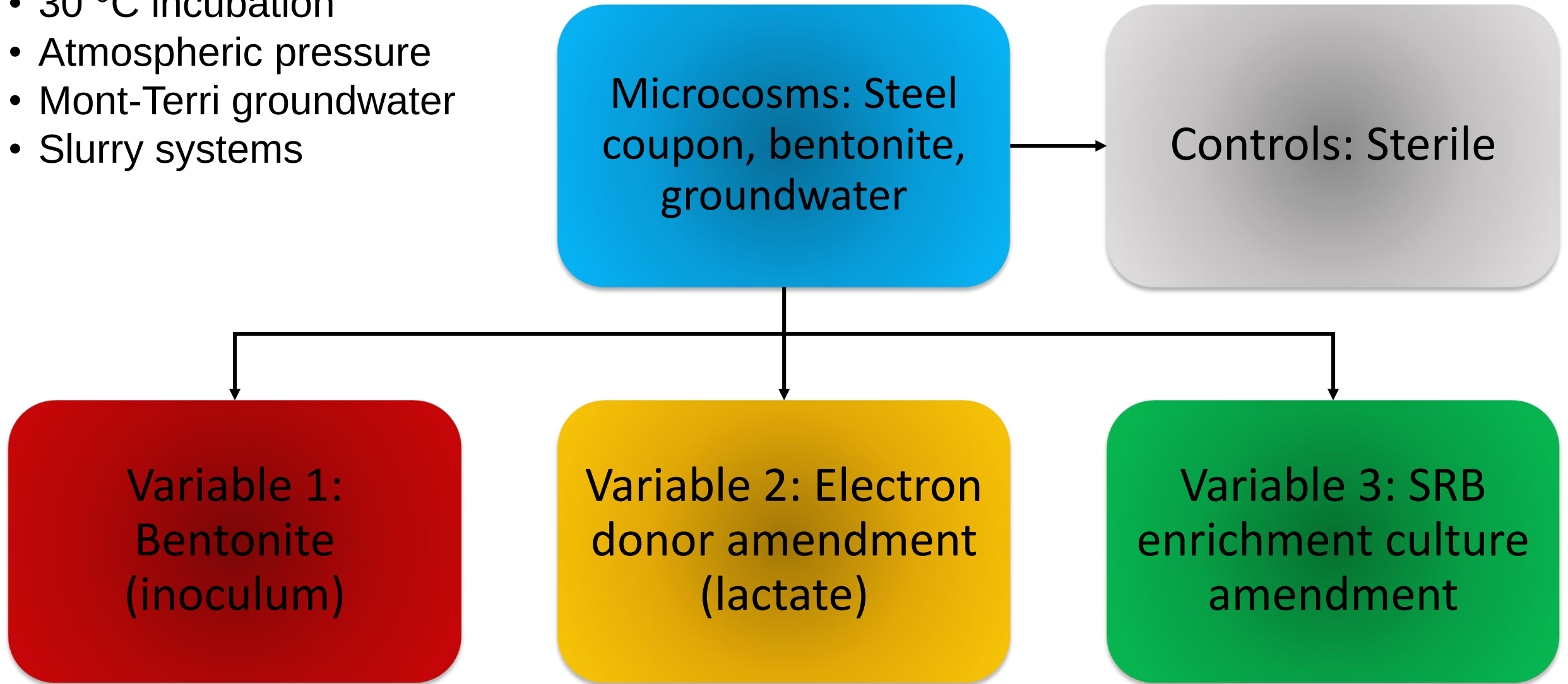
AIM:

*To evaluate
different
control
measures in
SRB-influenced
biocorrosion of
carbon steel*

Enriching native bentonite SRB communities



- 30 °C incubation
- Atmospheric pressure
- Mont-Terri groundwater
- Slurry systems



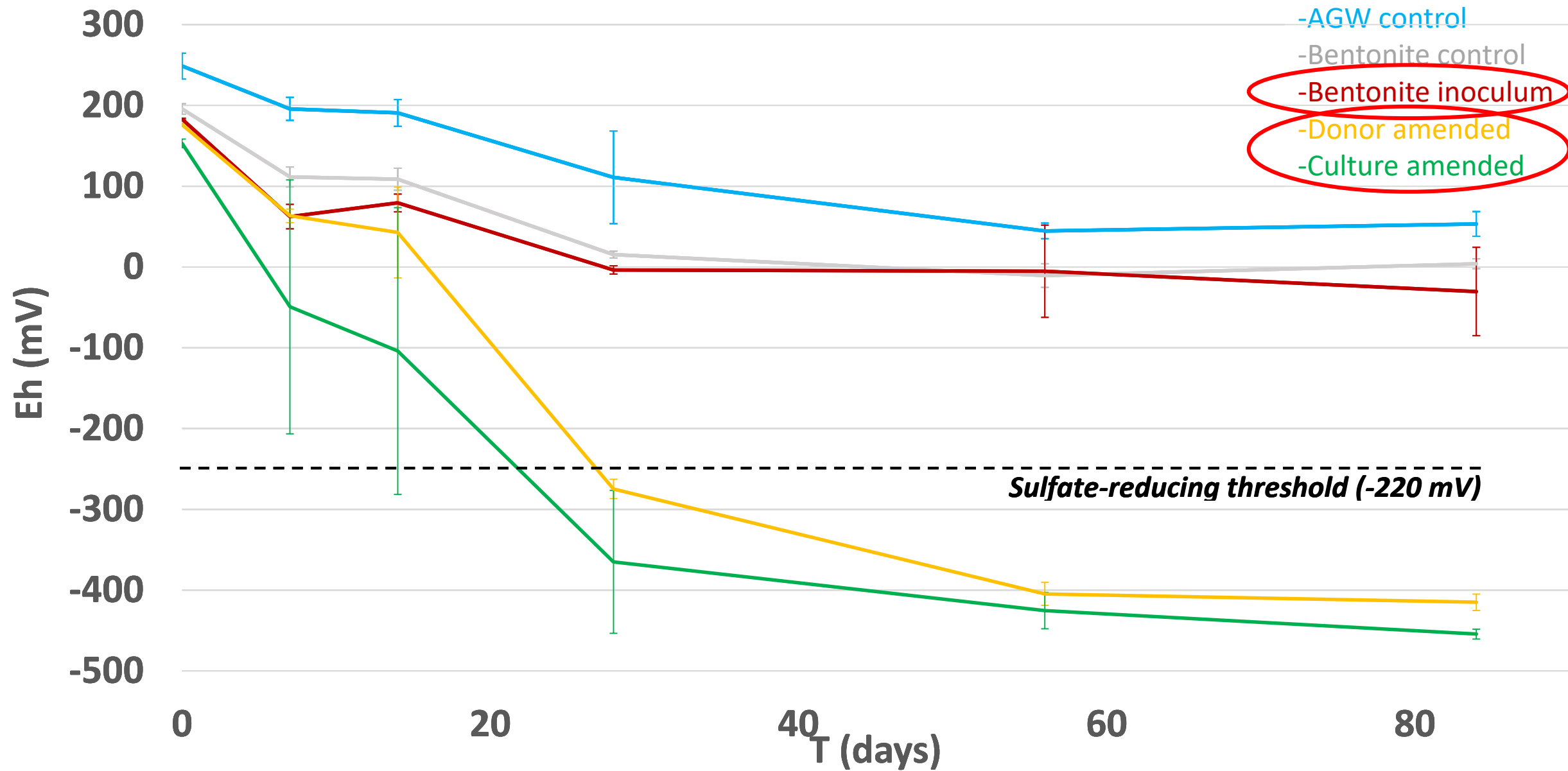
Microcosm
systems (40
ml):

Control

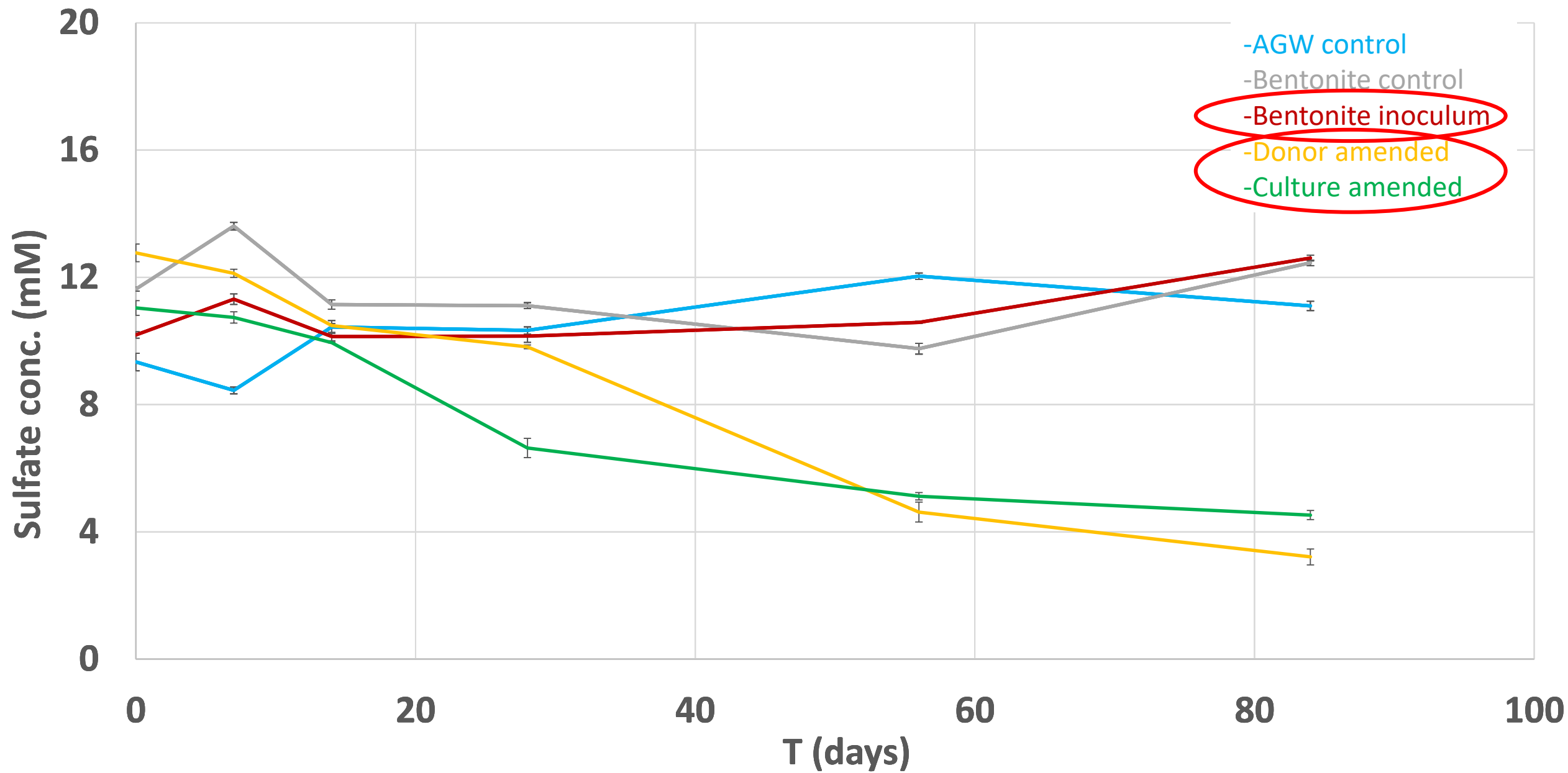
~75 mm

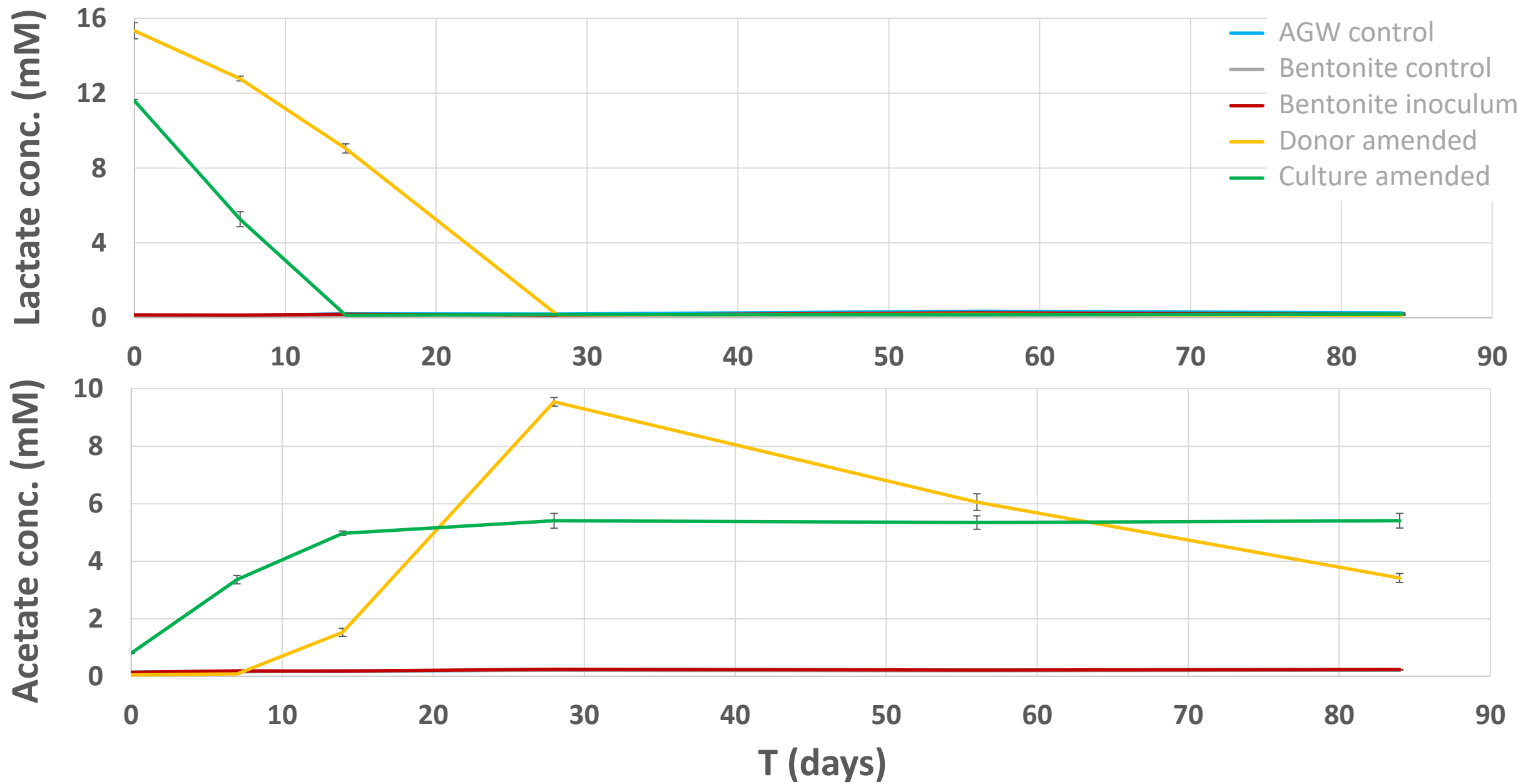
Steel
control

Full slurry

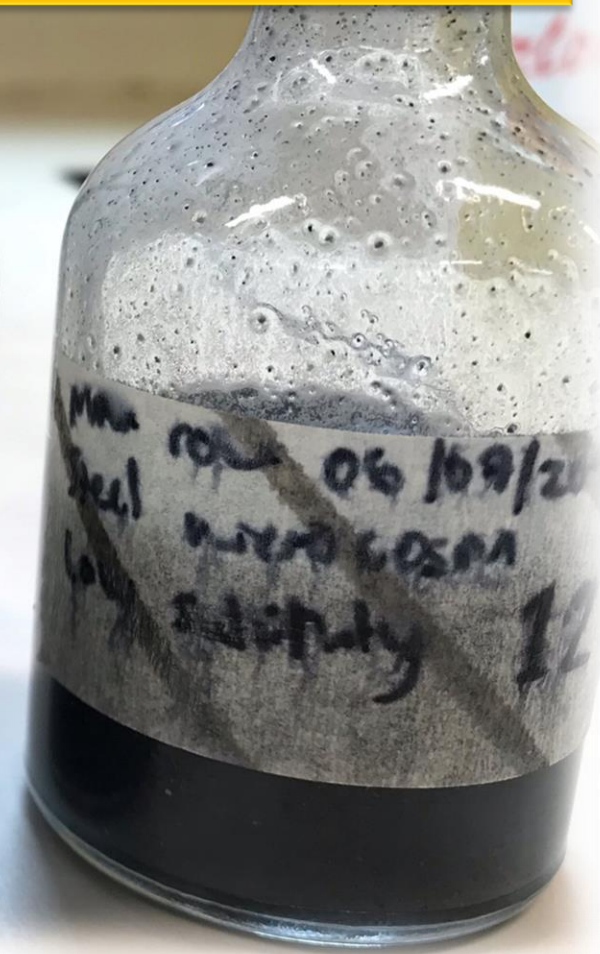
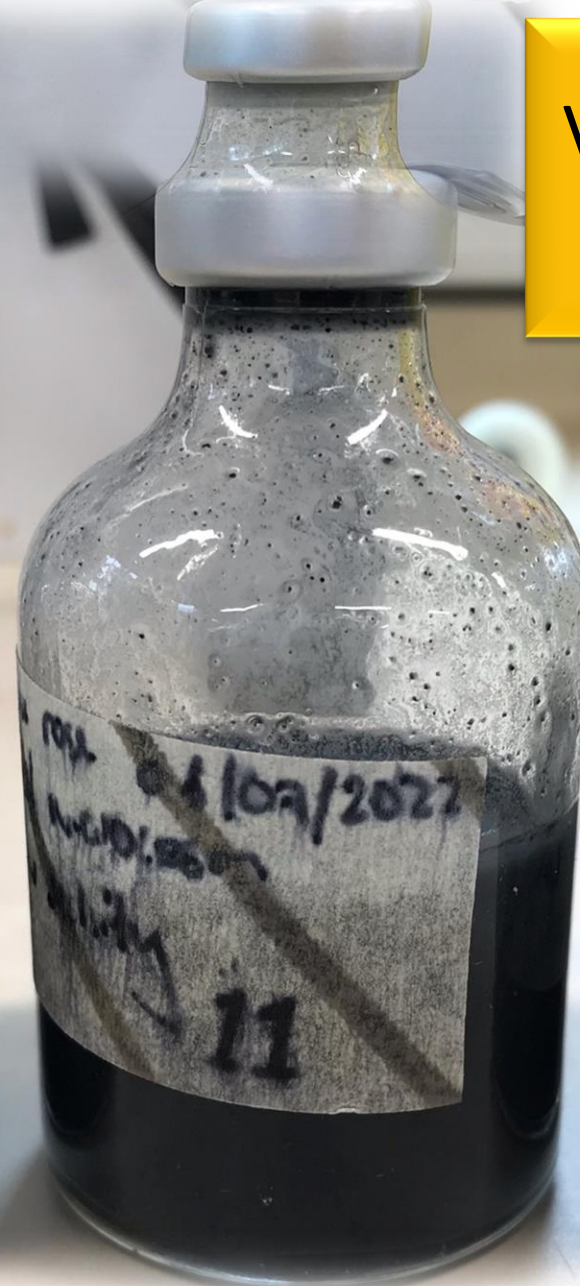
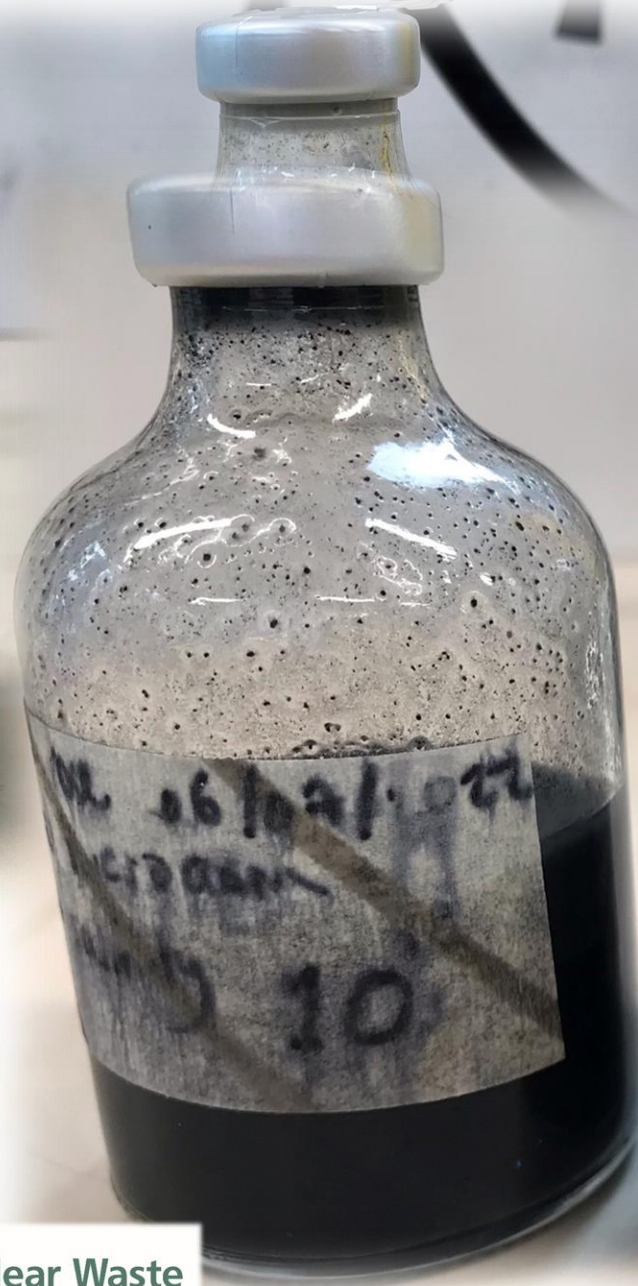
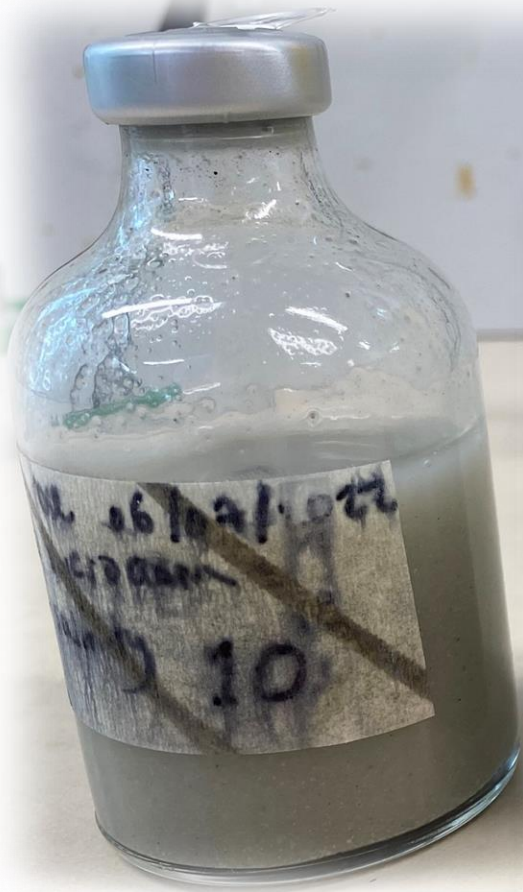


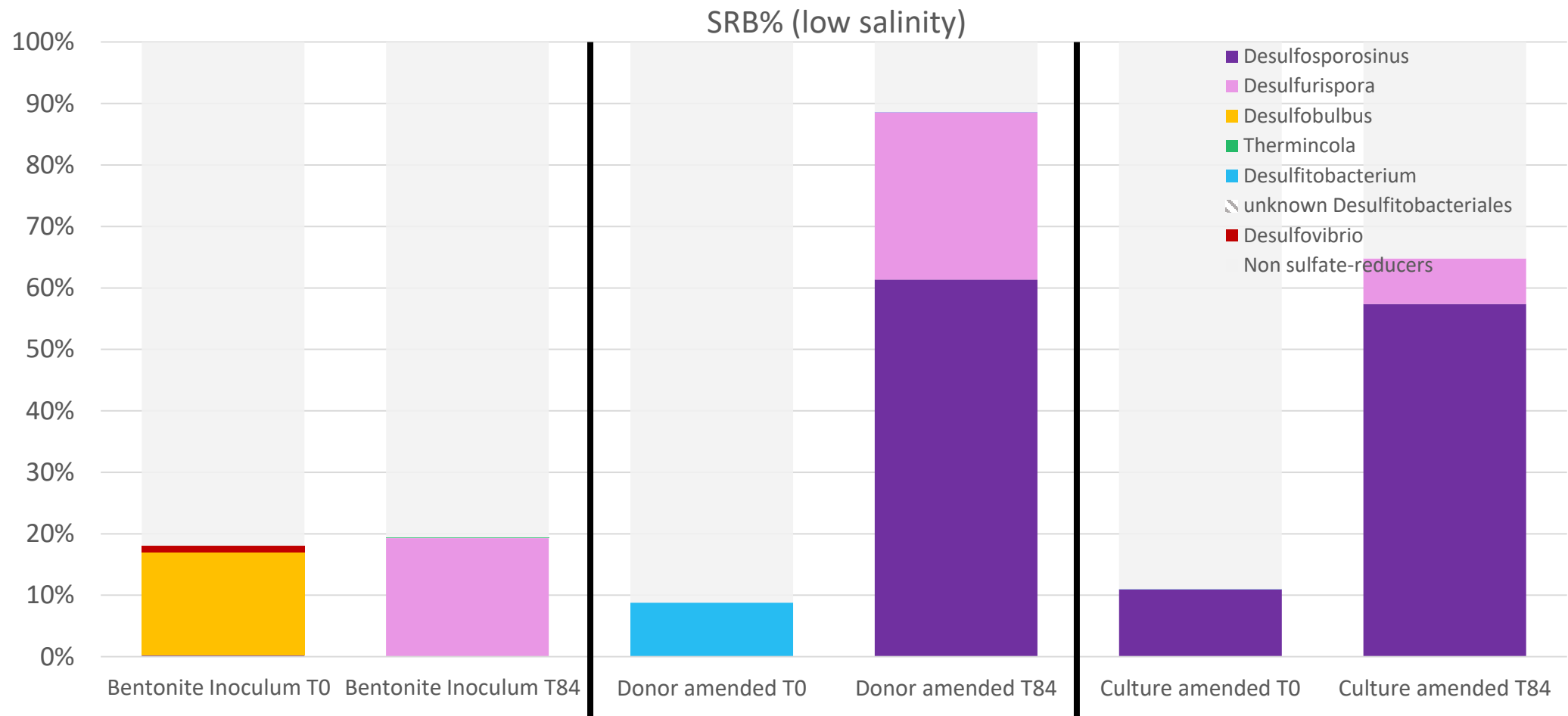
*autoclaved
C= carbon





Variable 2: Donor
amendment

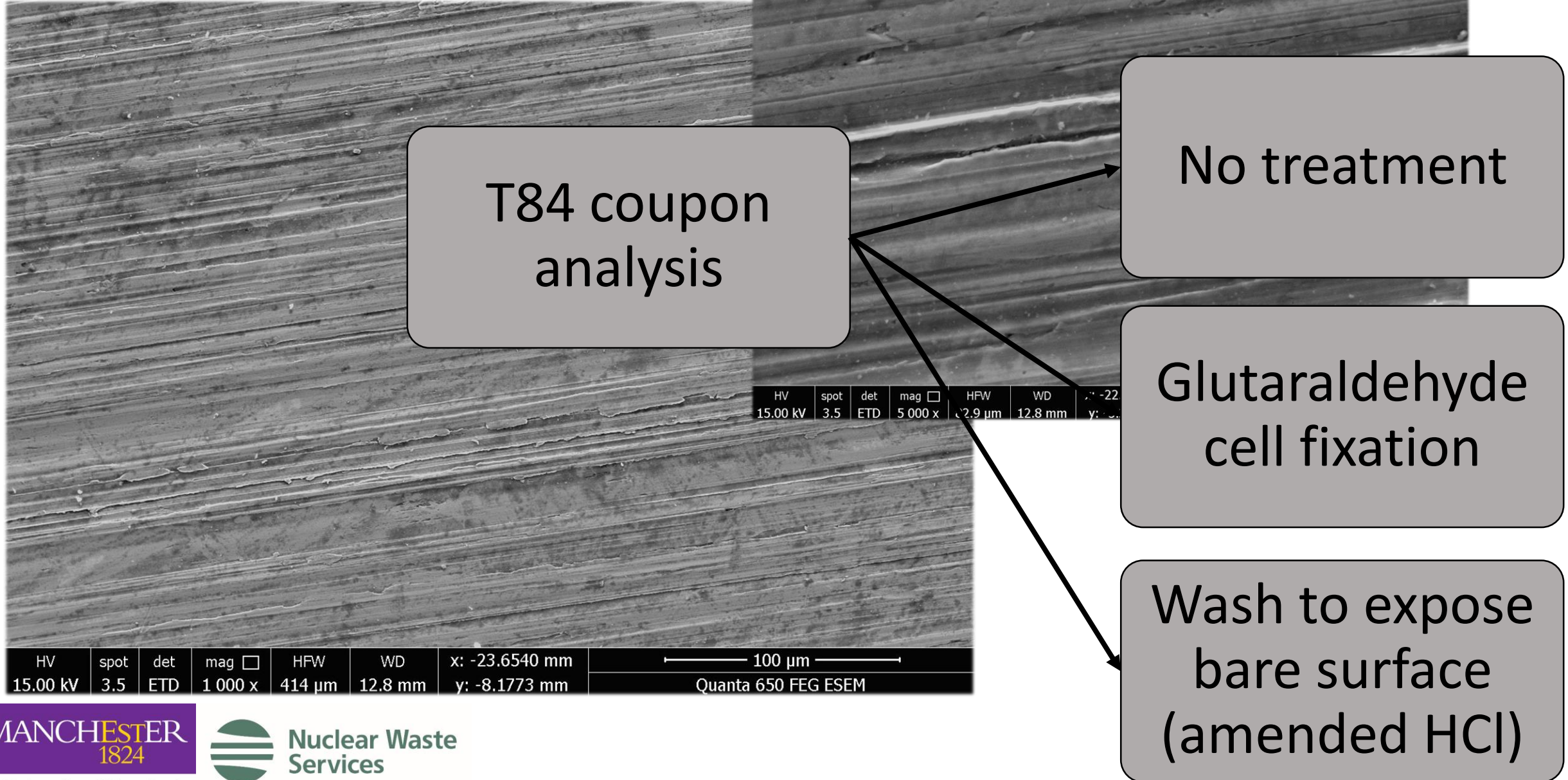




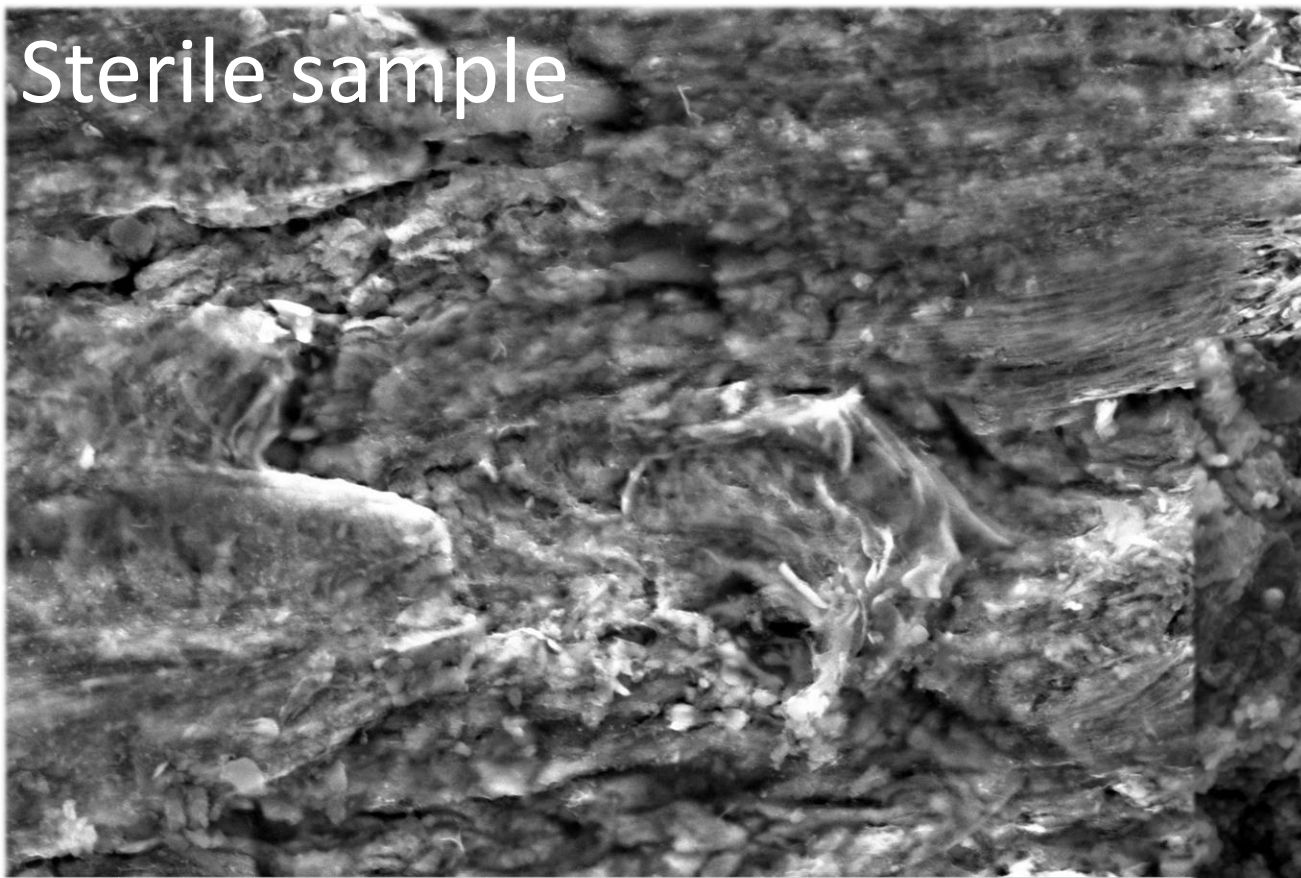
SRB cells/ gram sample

Sample	Cell numbers T0	Cell numbers T84
Bentonite	-	60,000
Donor amended	-	2,177,000
Culture amended	138,000	269,000

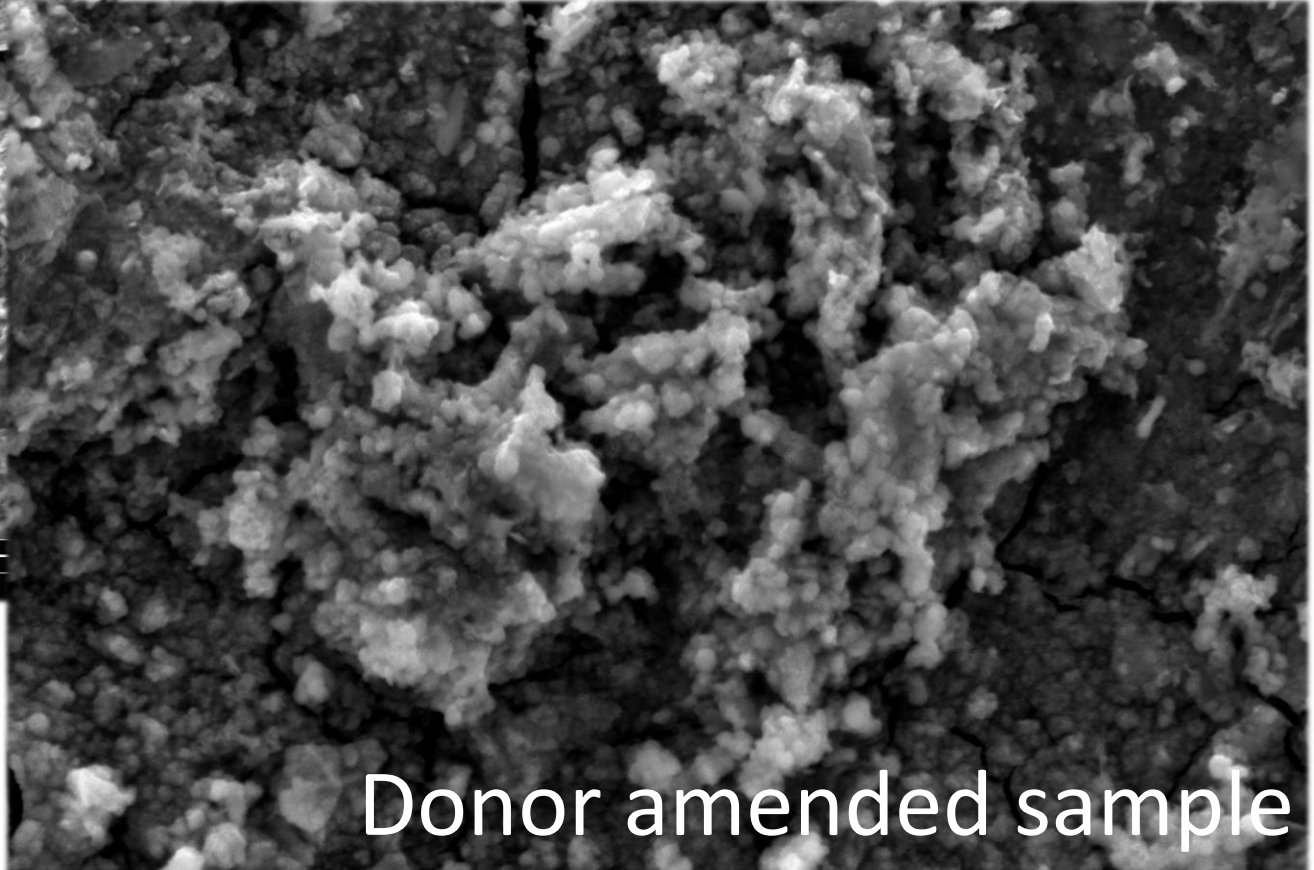
T0 steel surface



Sterile sample

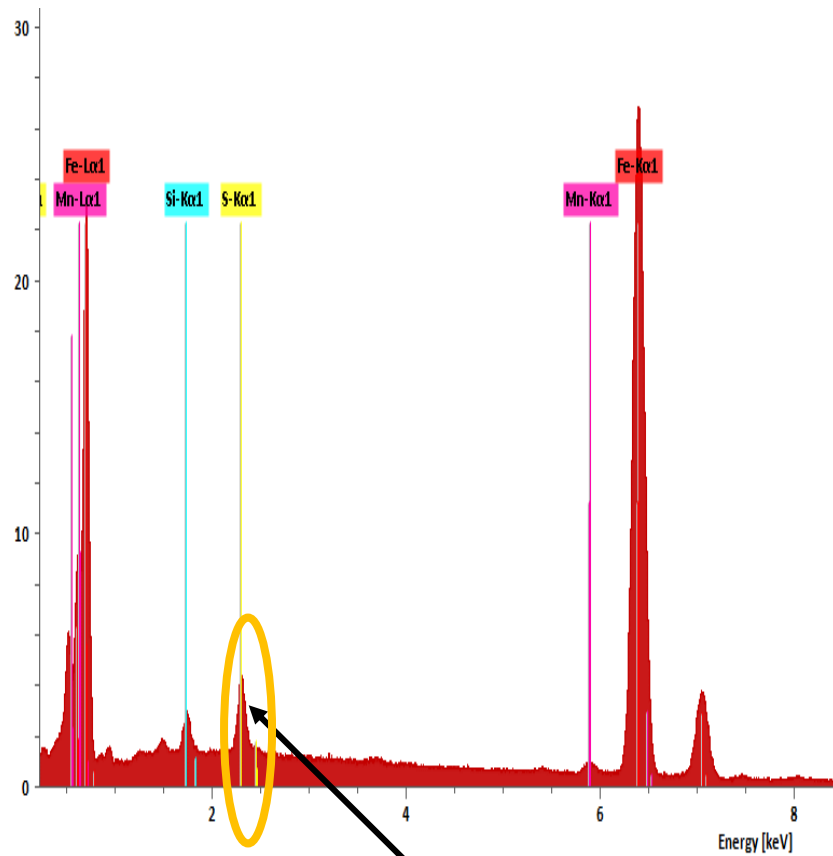


HV	spot	det	mag	□	HFW	WD	x: 1.6148 mm	10 µm
15.00 kV	3.0	ETD	13 115 x		31.6 µm	10.0 mm	y: -26.1323 mm	Quanta 650 FEG ESEM



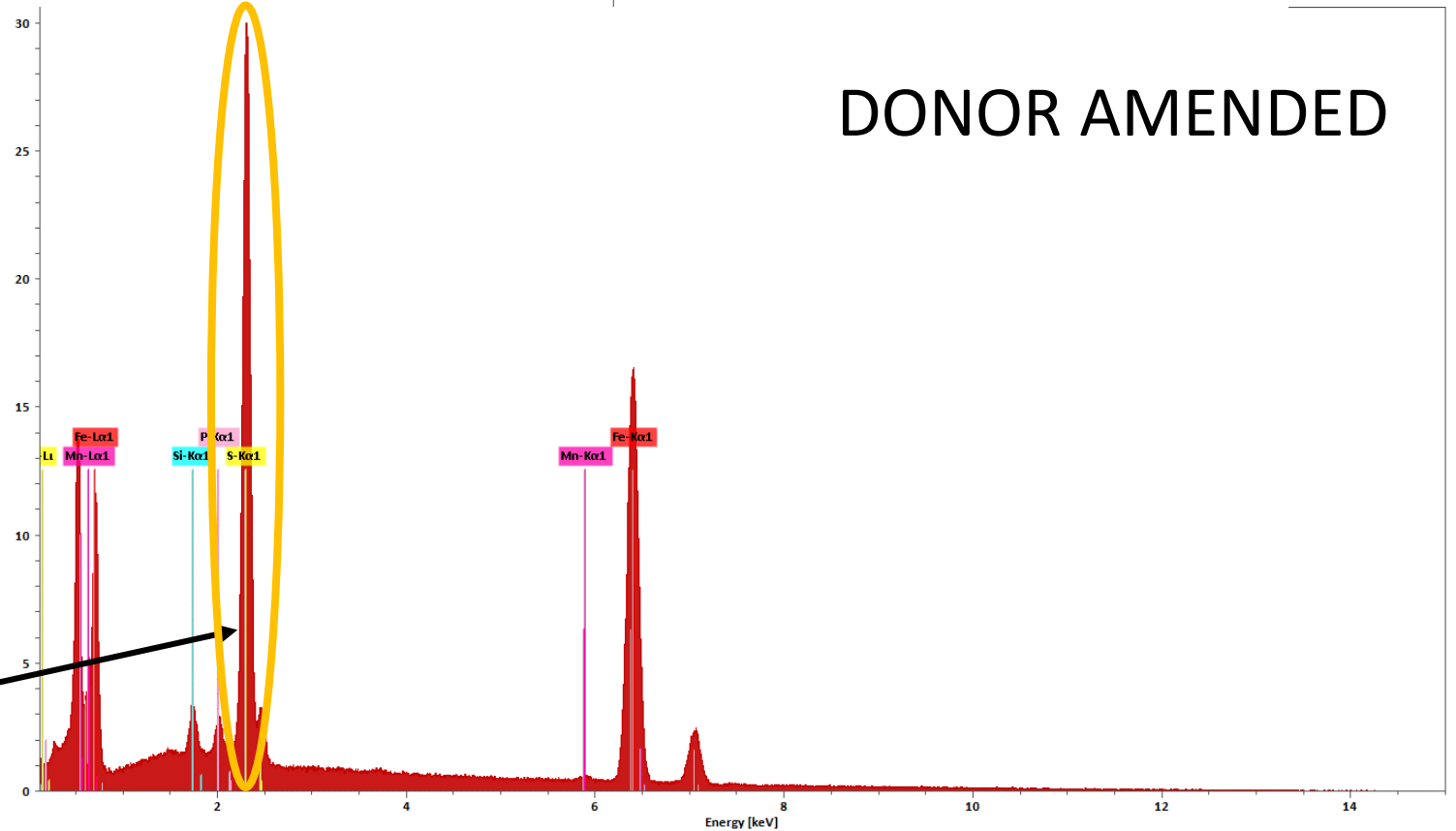
Donor amended sample

HV	spot	det	mag	□	HFW	WD	x: -5.6003 mm	10 µm
15.00 kV	3.0	ETD	12 482 x		33.2 µm	10.1 mm	y: 6.4773 mm	Quanta 650 FEG ESEM



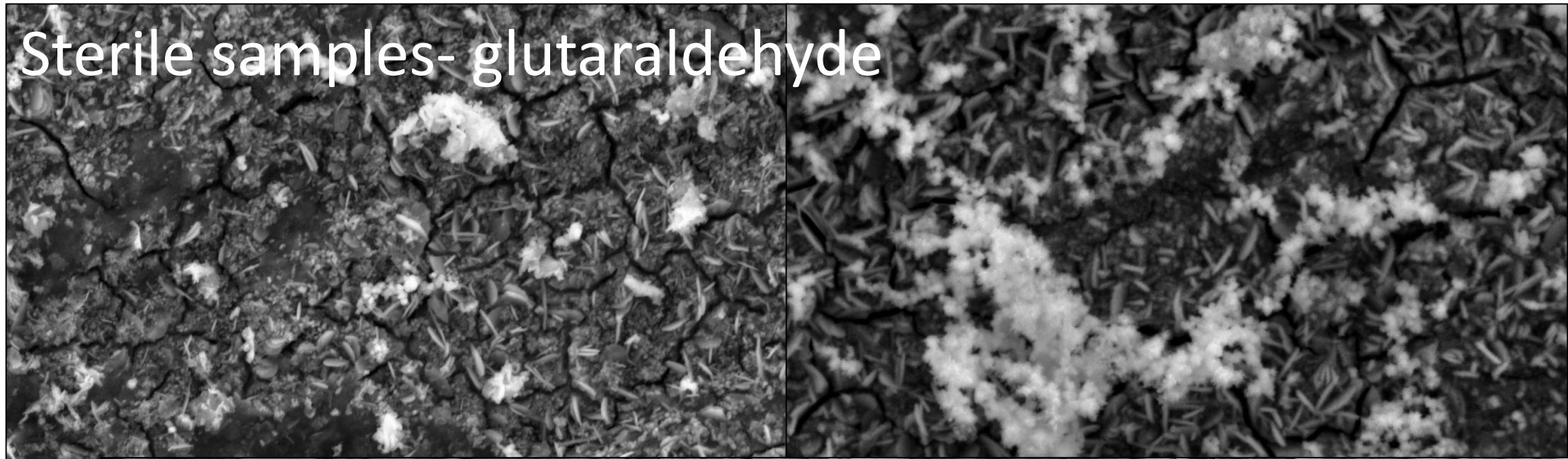
STERILE

DONOR AMENDED

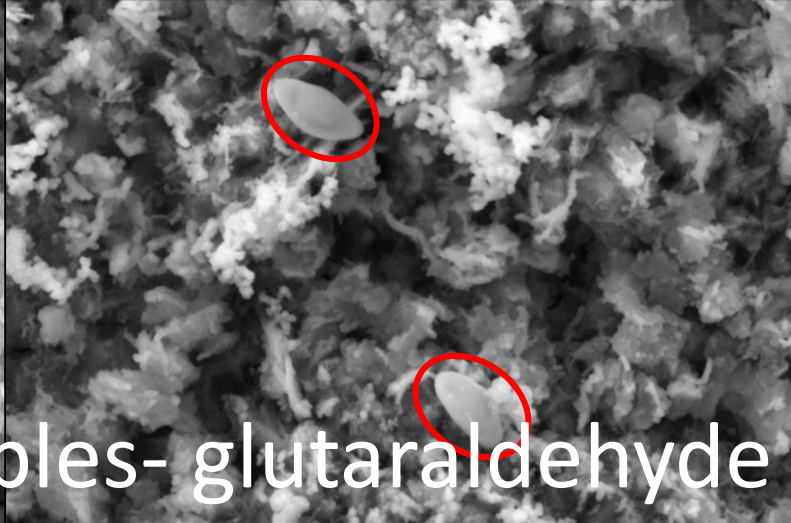
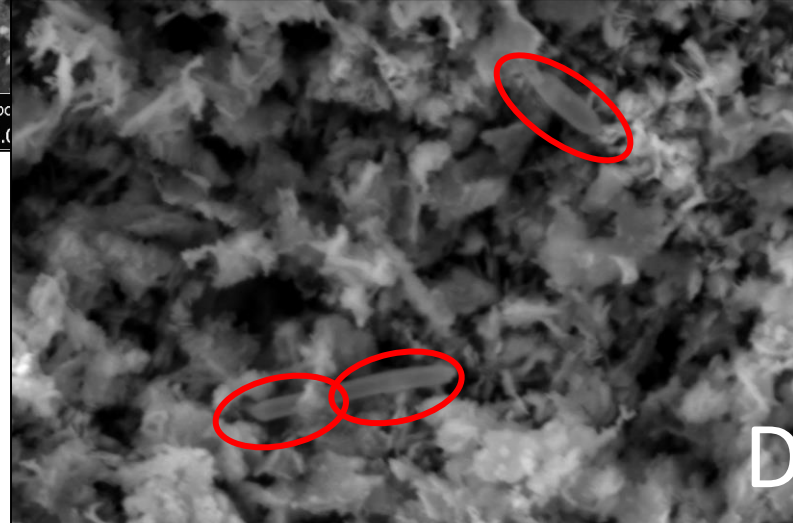


Sulfur

Sterile samples- glutaraldehyde



HV 10.00 kV spot 3.0

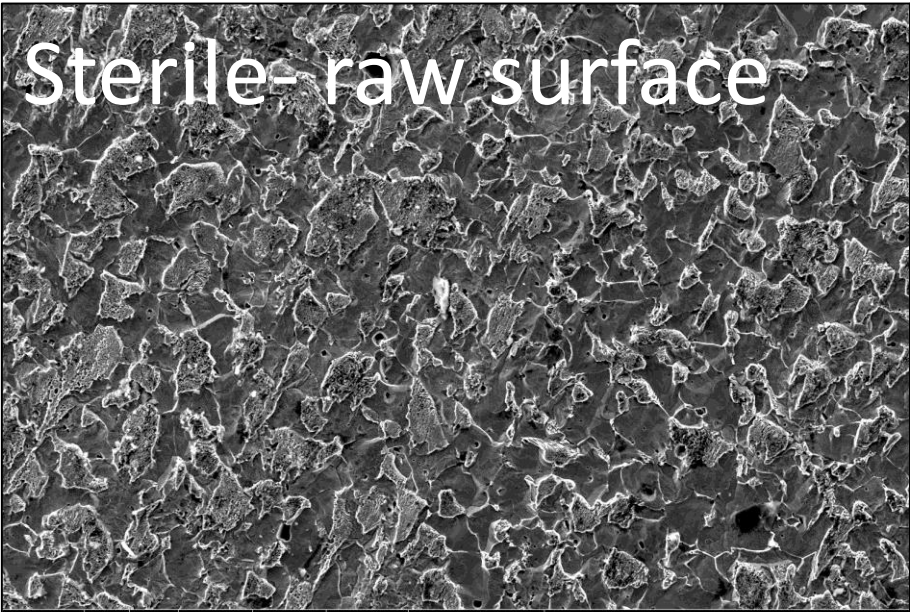


Donor amended samples- glutaraldehyde

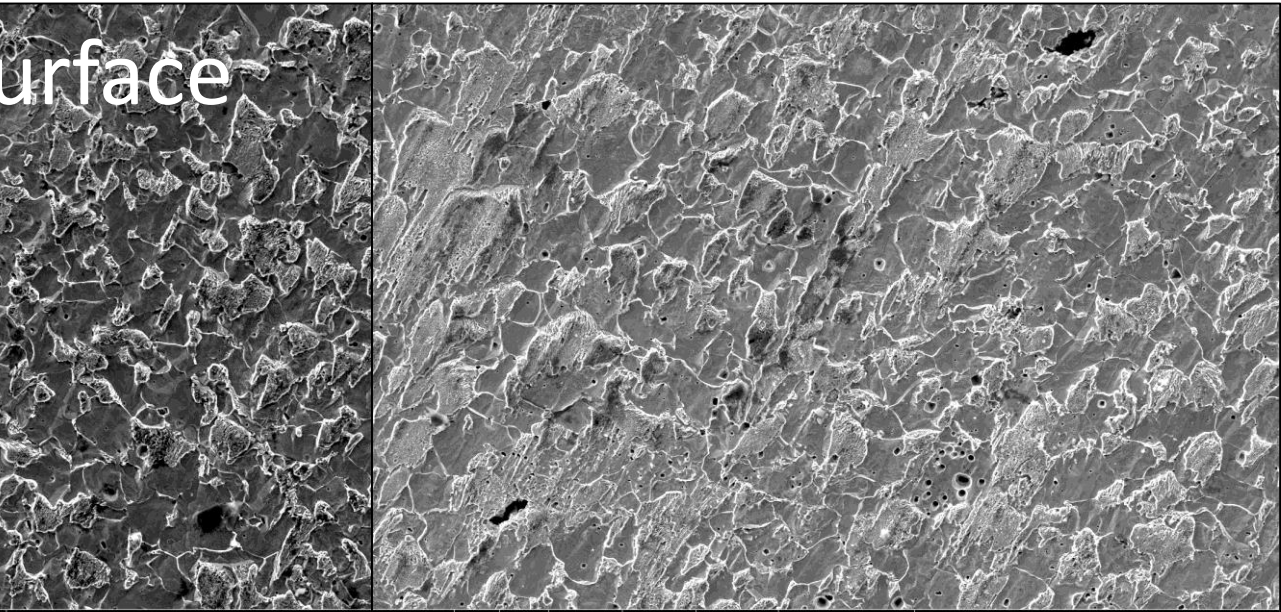
HV 10.00 kV spot 3.0 det ETD mag 17 427 x HFW 23.8 µm WD 10.1 mm x: -11.4723 mm y: -7.4708 mm 5 µm Quanta 650 FEG ESEM

HV 10.00 kV spot 3.0 det ETD mag 13 579 x HFW 30.5 µm WD 10.1 mm x: -6.7235 mm y: -12.4190 mm 10 µm Quanta 650 FEG ESEM

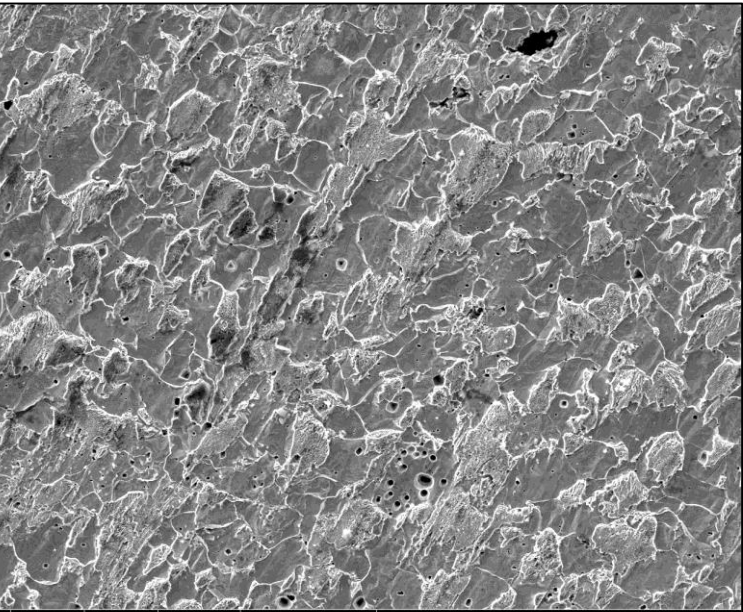
HV 10.00 kV spot 3.0 det ETD mag 20 964 x HFW 19.8 µm WD 10.1 mm x: -10.2863 mm y: -11.5985 mm 5 µm Quanta 650 FEG ESEM



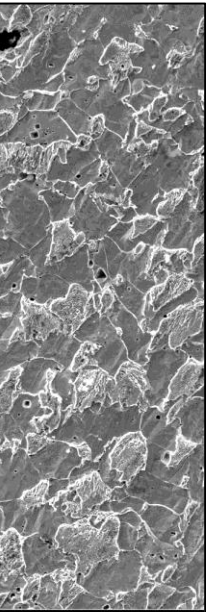
HV	spot	det	mag	□	HFV	WD	x:	y:
15.00 kV	3.5	ETD	1 252 x		331 μm	10.7 mm	10.8598 mm	19.3680 mm



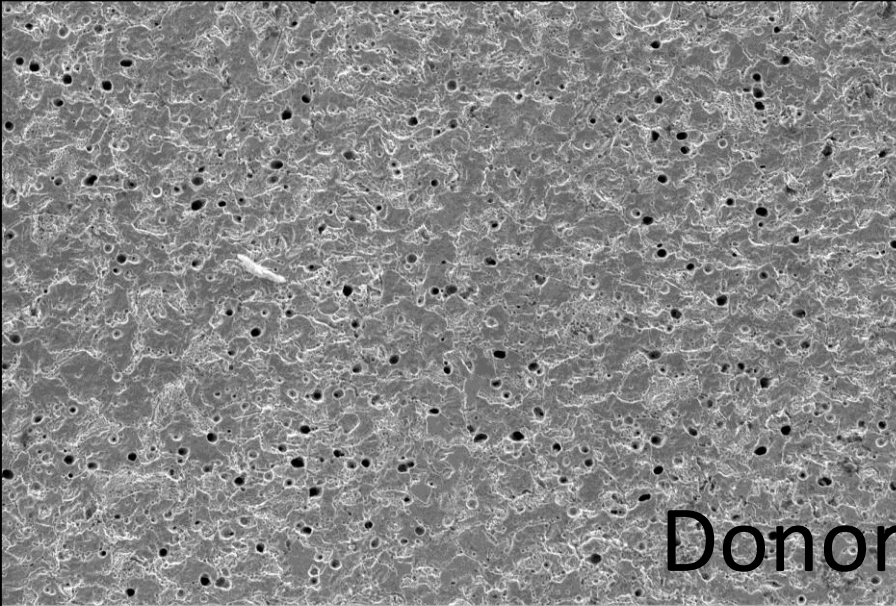
HV	spot	det	mag	□	HFV	WD	x:	y:
15.00 kV	3.5	ETD	1 081 x		383 μm	10.7 mm	7.4950 mm	19.3313 mm



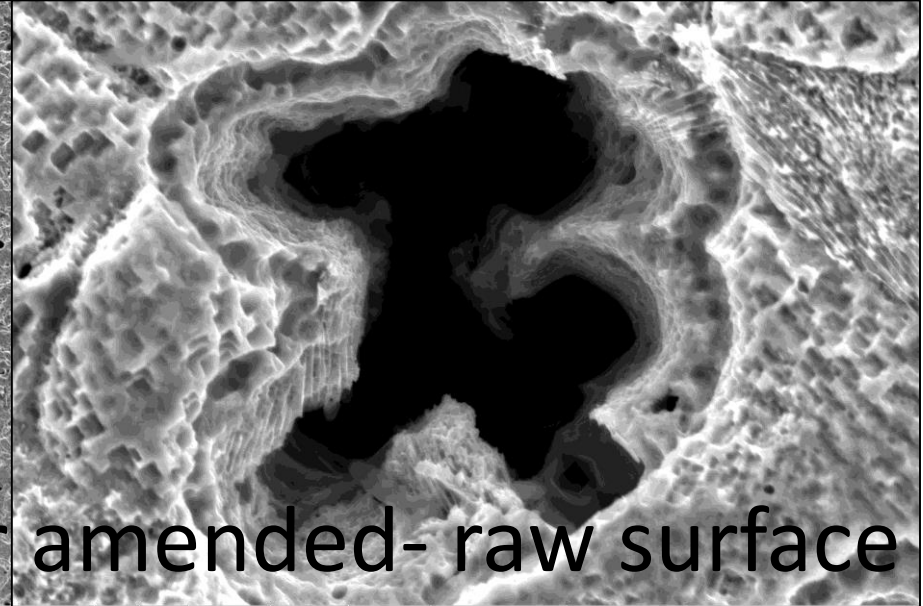
HV	spot	det	mag	□	HFV	WD	x:	y:
15.00 kV	3.5	ETD	1 081 x		383 μm	10.7 mm	7.4950 mm	19.3313 mm



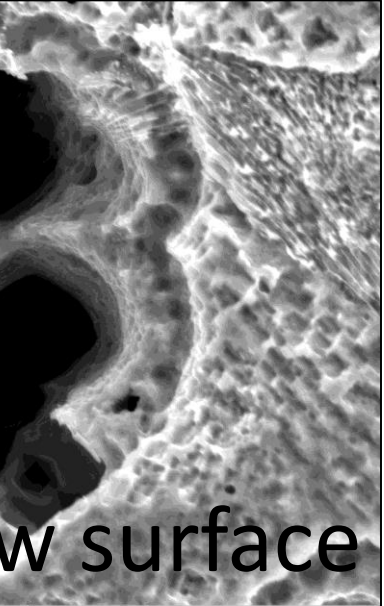
HV	spot	det	mag	□	HFV	WD	x:	y:
15.00 kV	3.5	ETD	1 081 x		383 μm	10.7 mm	7.4950 mm	19.3313 mm



HV	spot	det	mag	□	HFV	WD	x:	y:
15.00 kV	3.5	ETD	852 x		486 μm	11.2 mm	20.9128 mm	-1.7938 mm



HV	spot	det	mag	□	HFV	WD	x:	y:
15.00 kV	3.5	ETD	13 629 x		30.4 μm	11.1 mm	27.9120 mm	-2.8048 mm

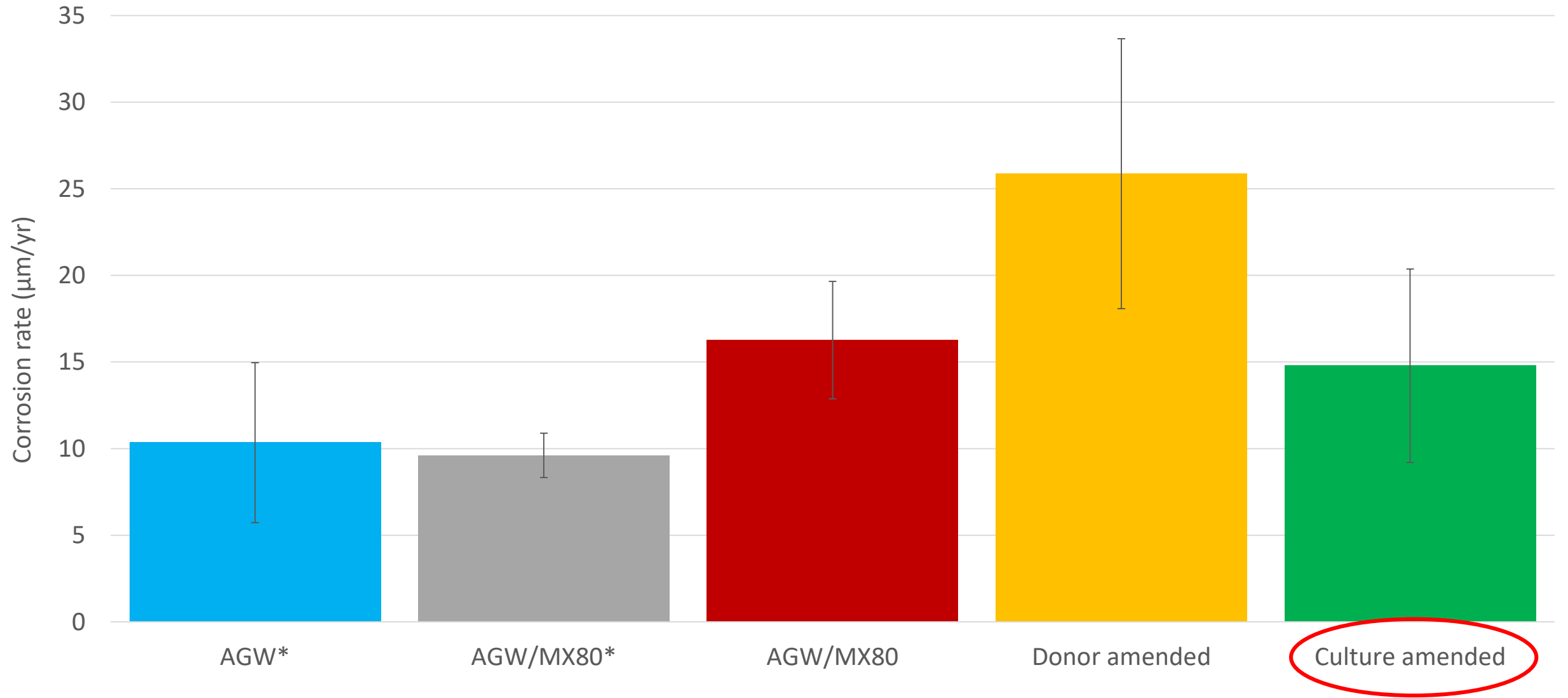


HV	spot	det	mag	□	HFV	WD	x:	y:
15.00 kV	3.5	ETD	13 629 x		30.4 μm	11.1 mm	27.9120 mm	-2.8048 mm

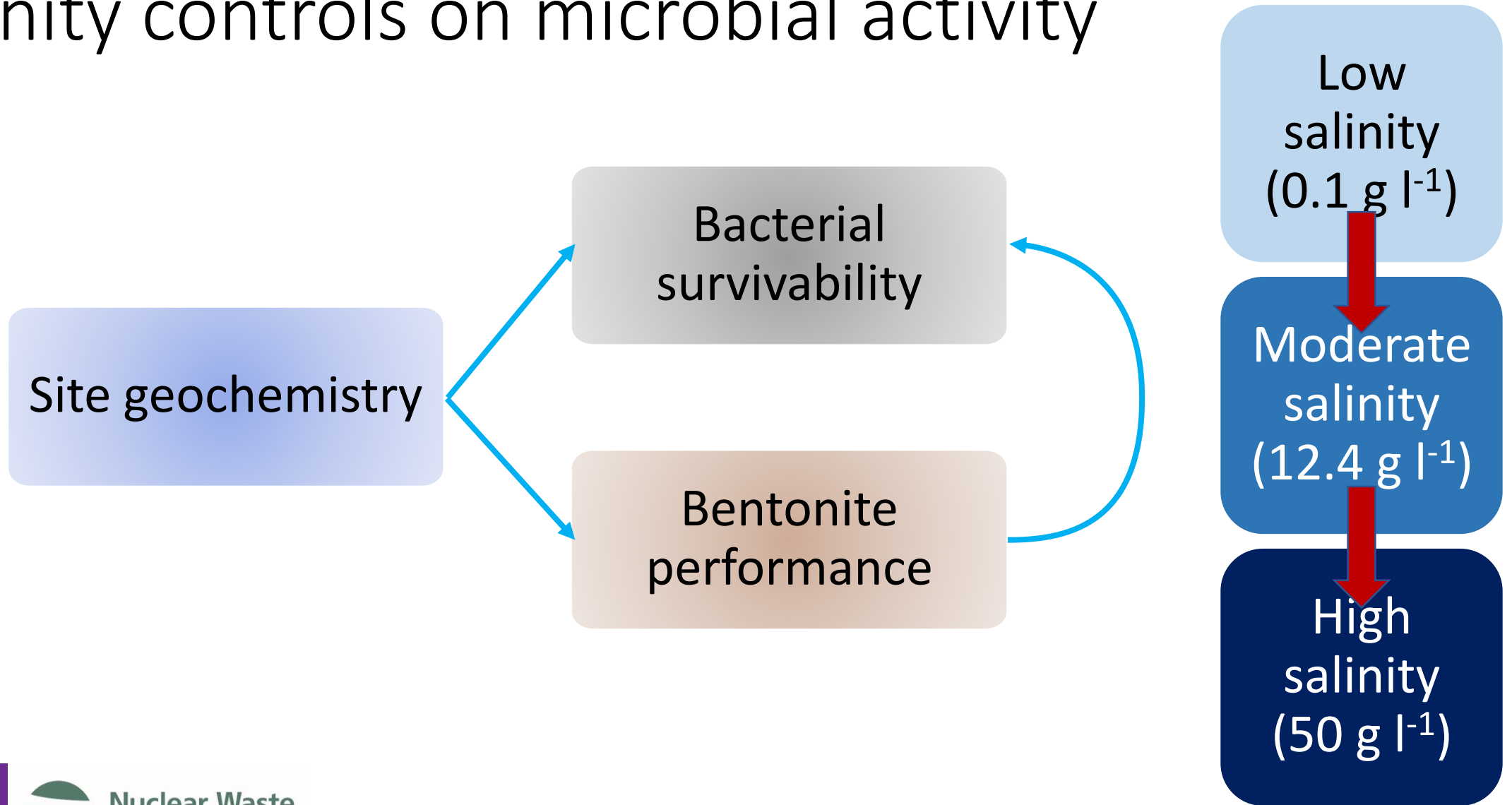
Sterile- raw surface

Donor amended- raw surface

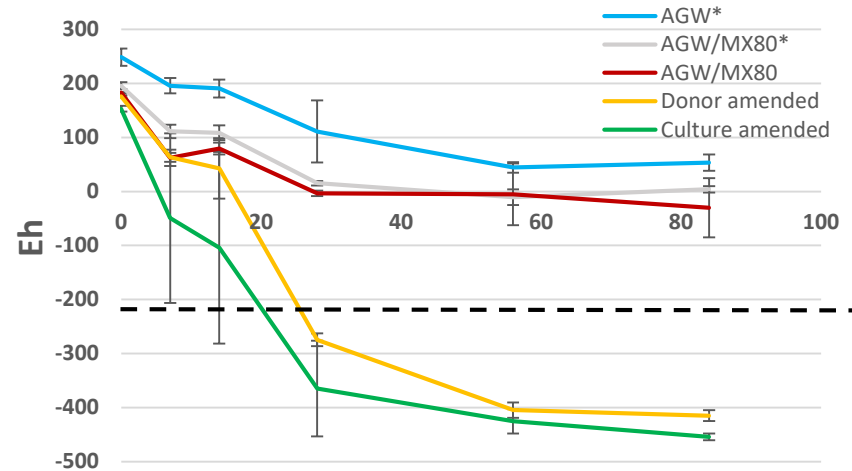
Corrosion rates from mass loss



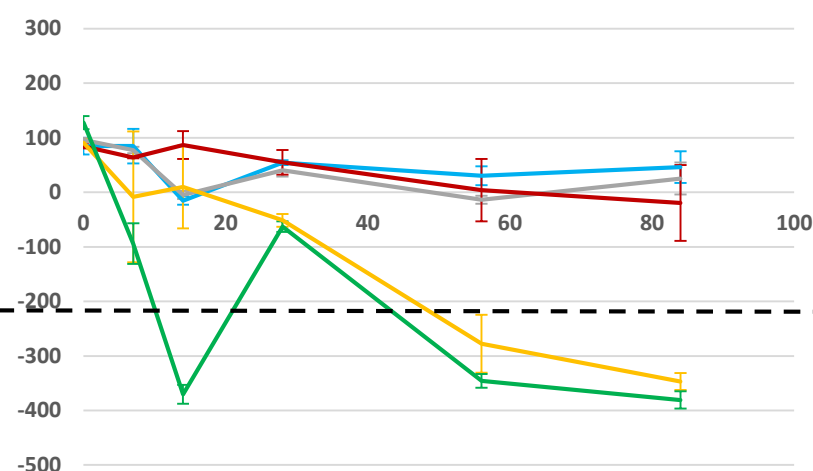
Salinity controls on microbial activity



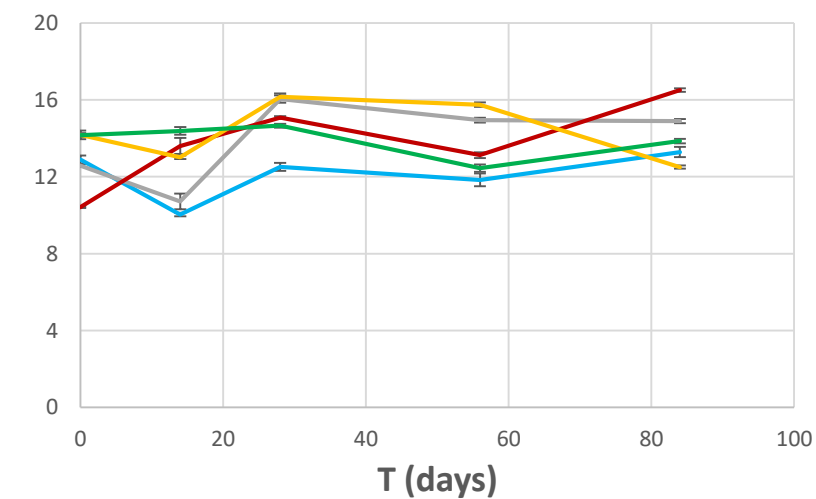
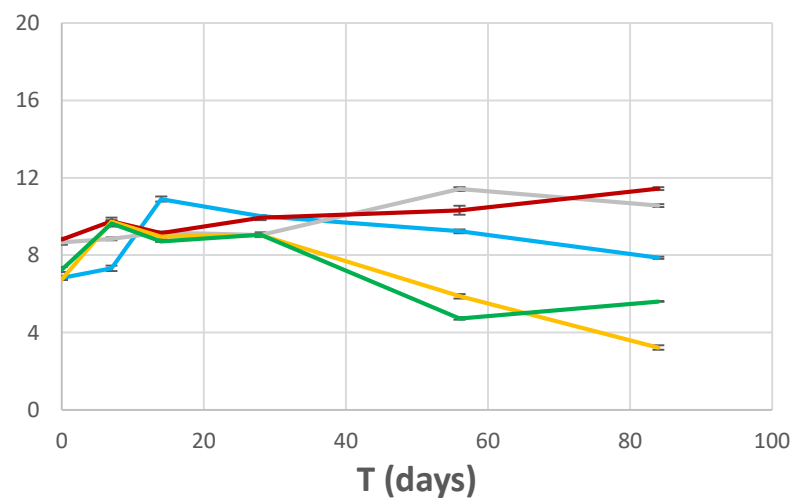
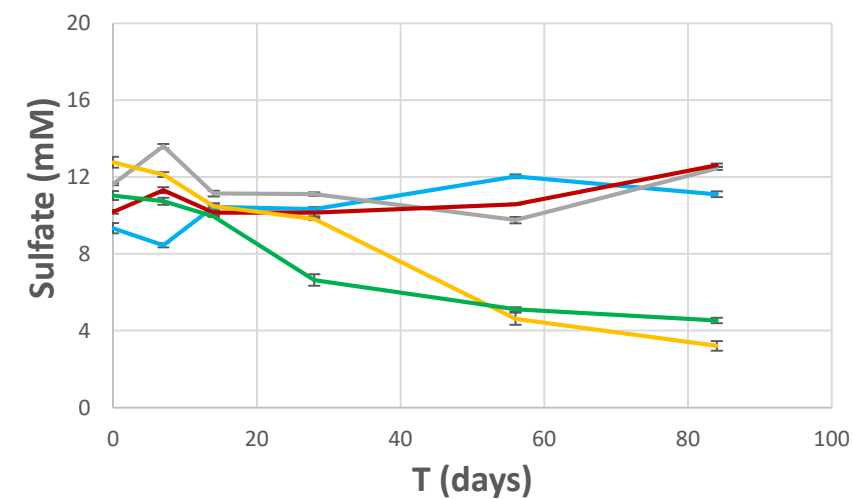
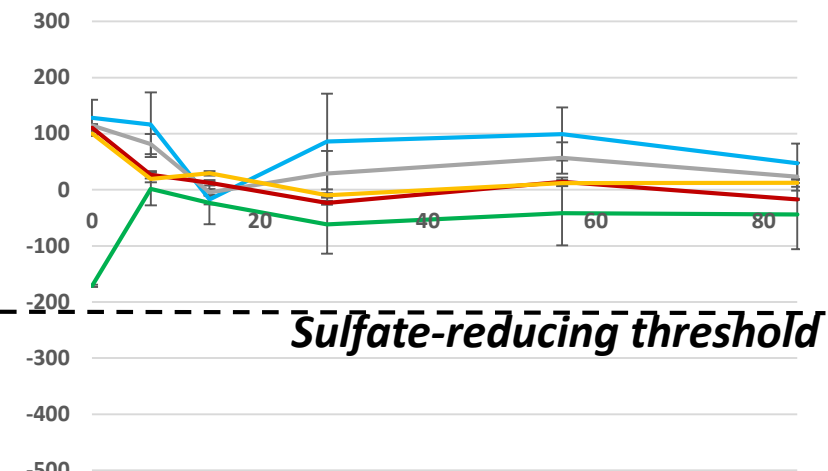
Low NaCl (0.1 g l⁻¹)



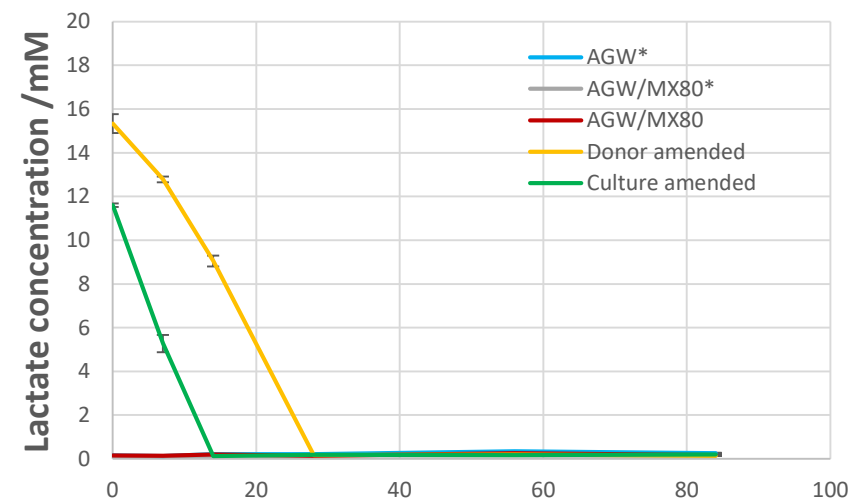
Moderate NaCl (12.4 g l⁻¹)



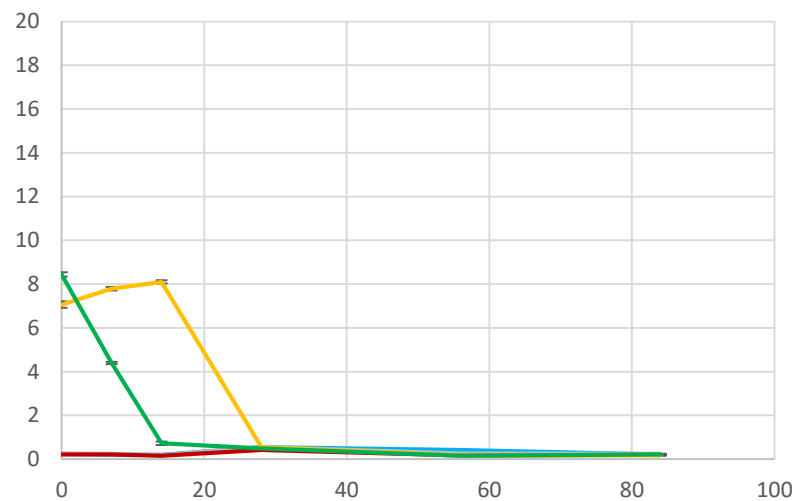
High NaCl (50 g l⁻¹)



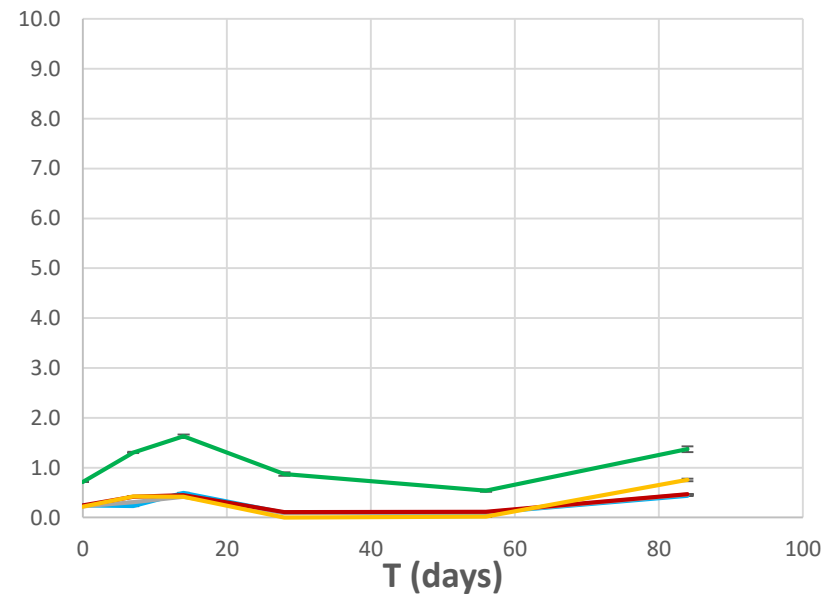
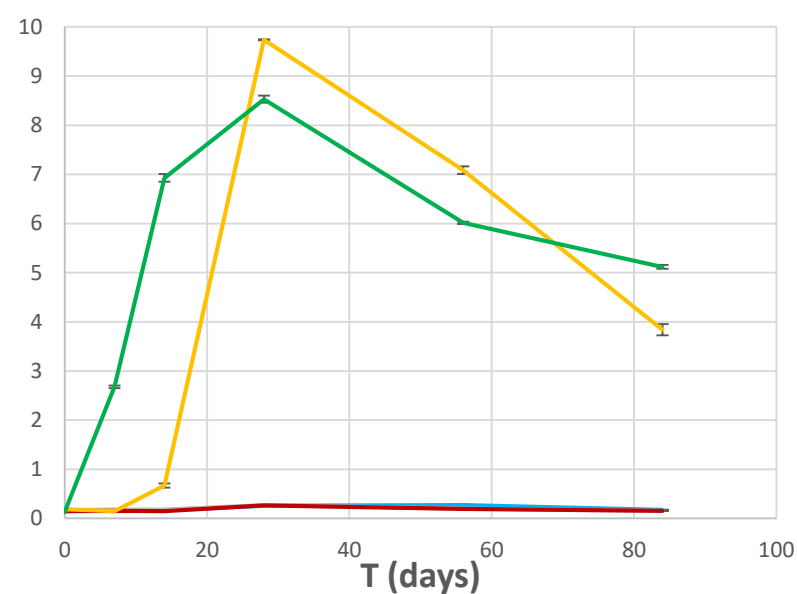
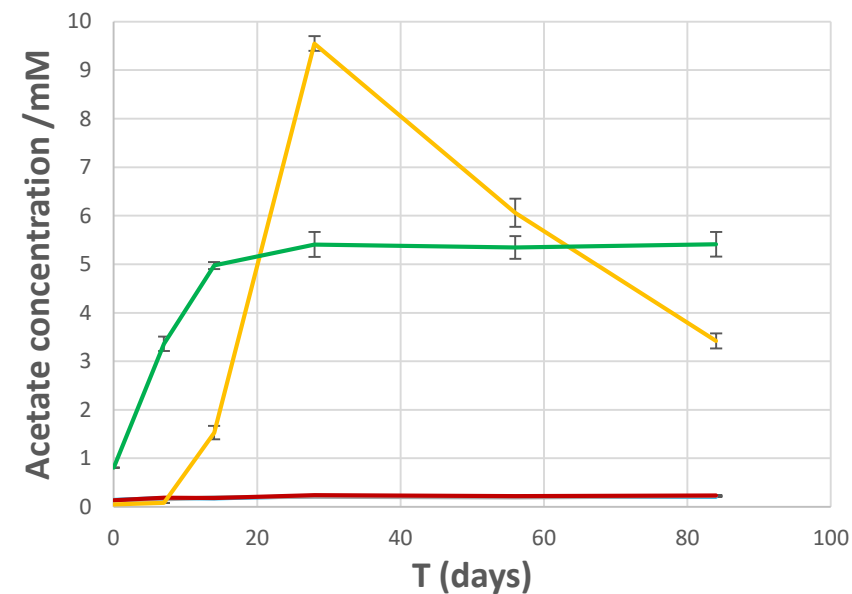
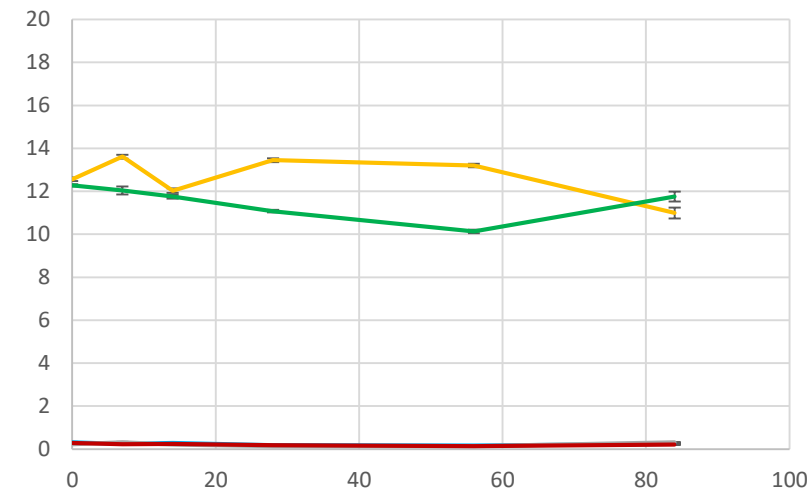
Low NaCl (0.1 g l⁻¹)



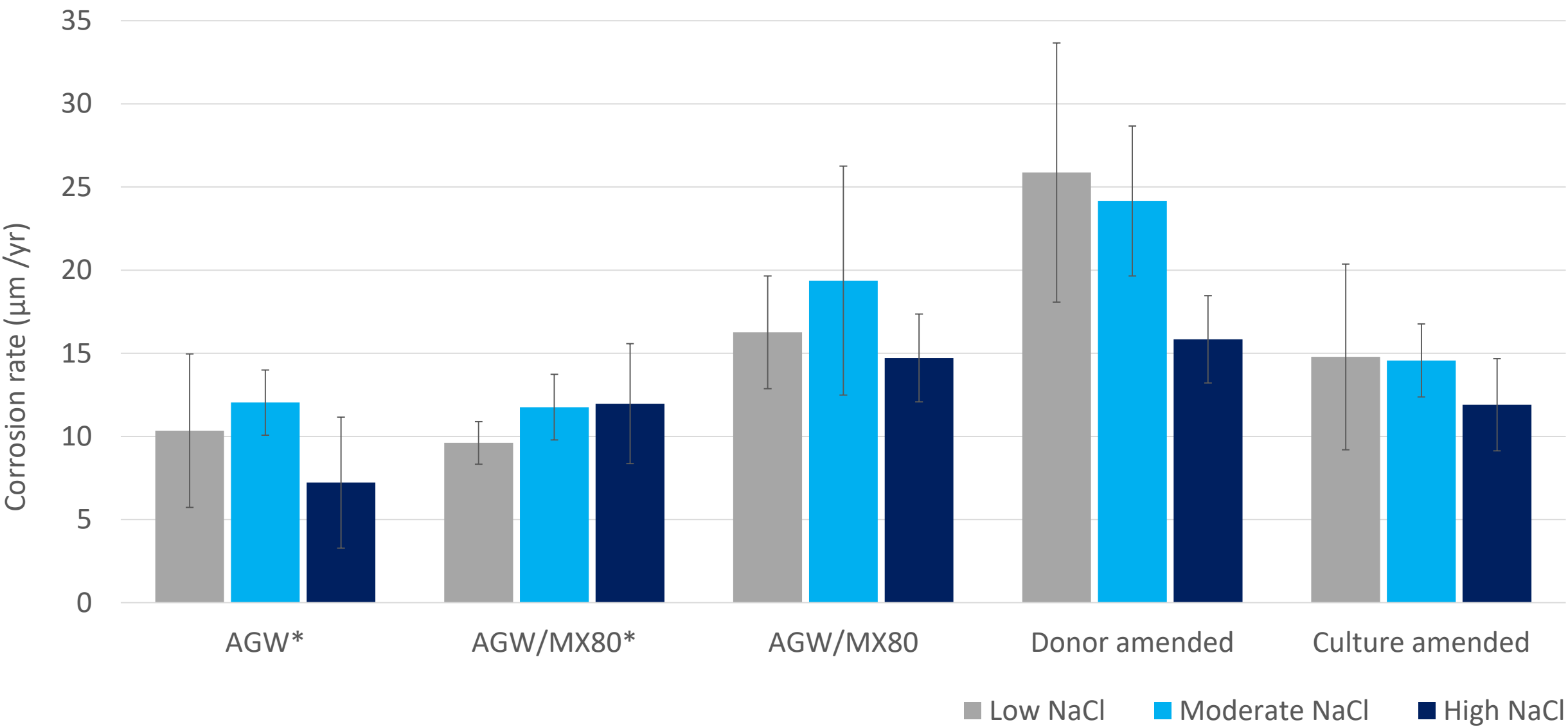
Moderate NaCl (12.4 g l⁻¹)



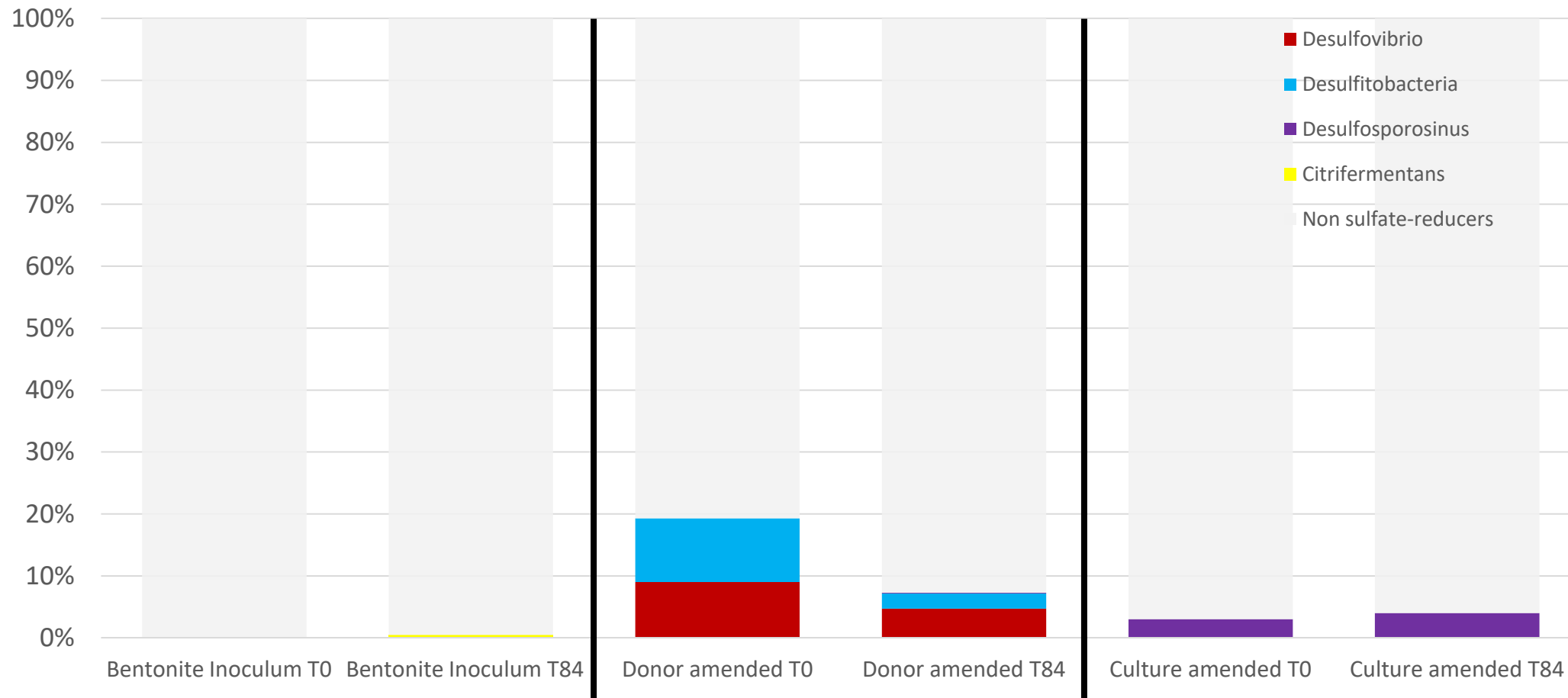
High NaCl (50 g l⁻¹)



Corrosion rates from mass loss

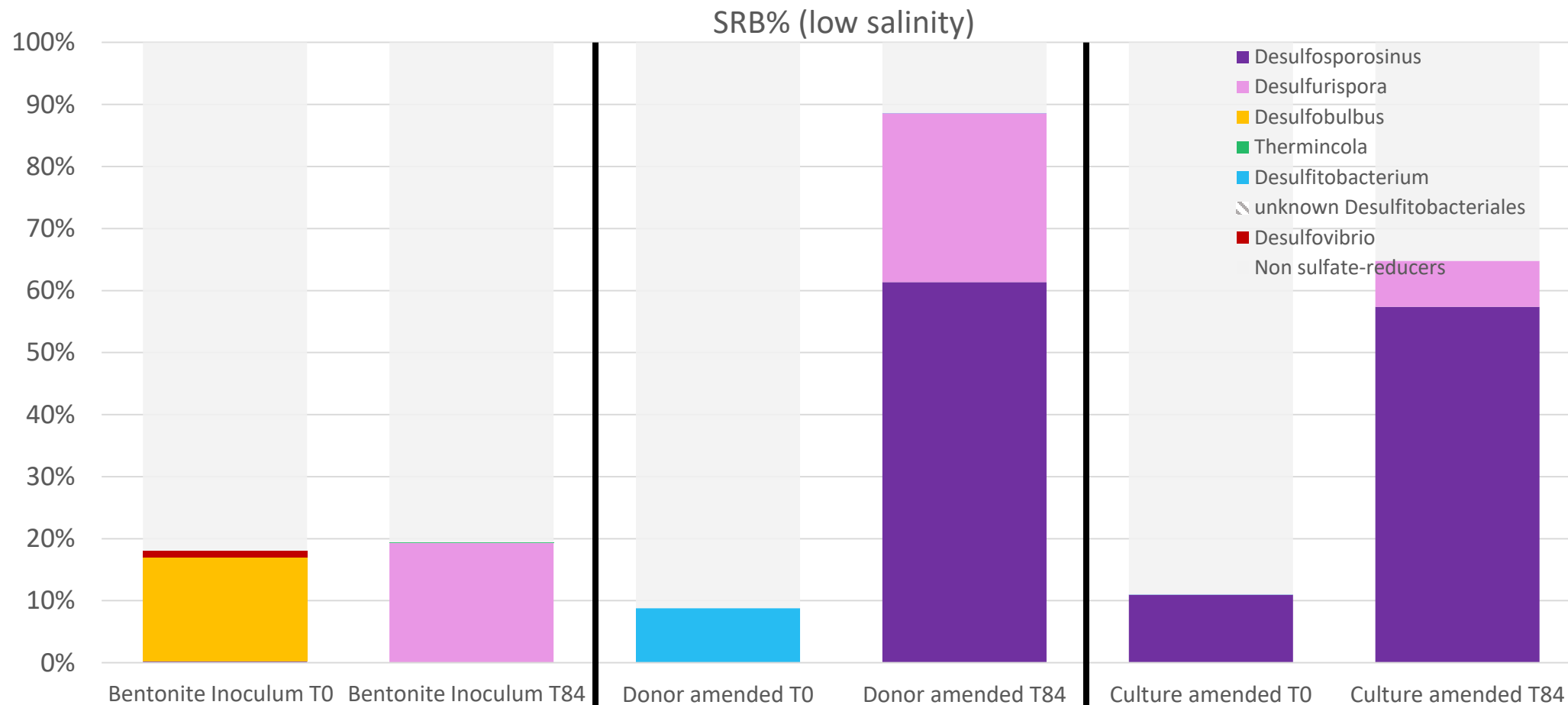


SRB% by genus (high salinity)



SRB cells/ gram sample

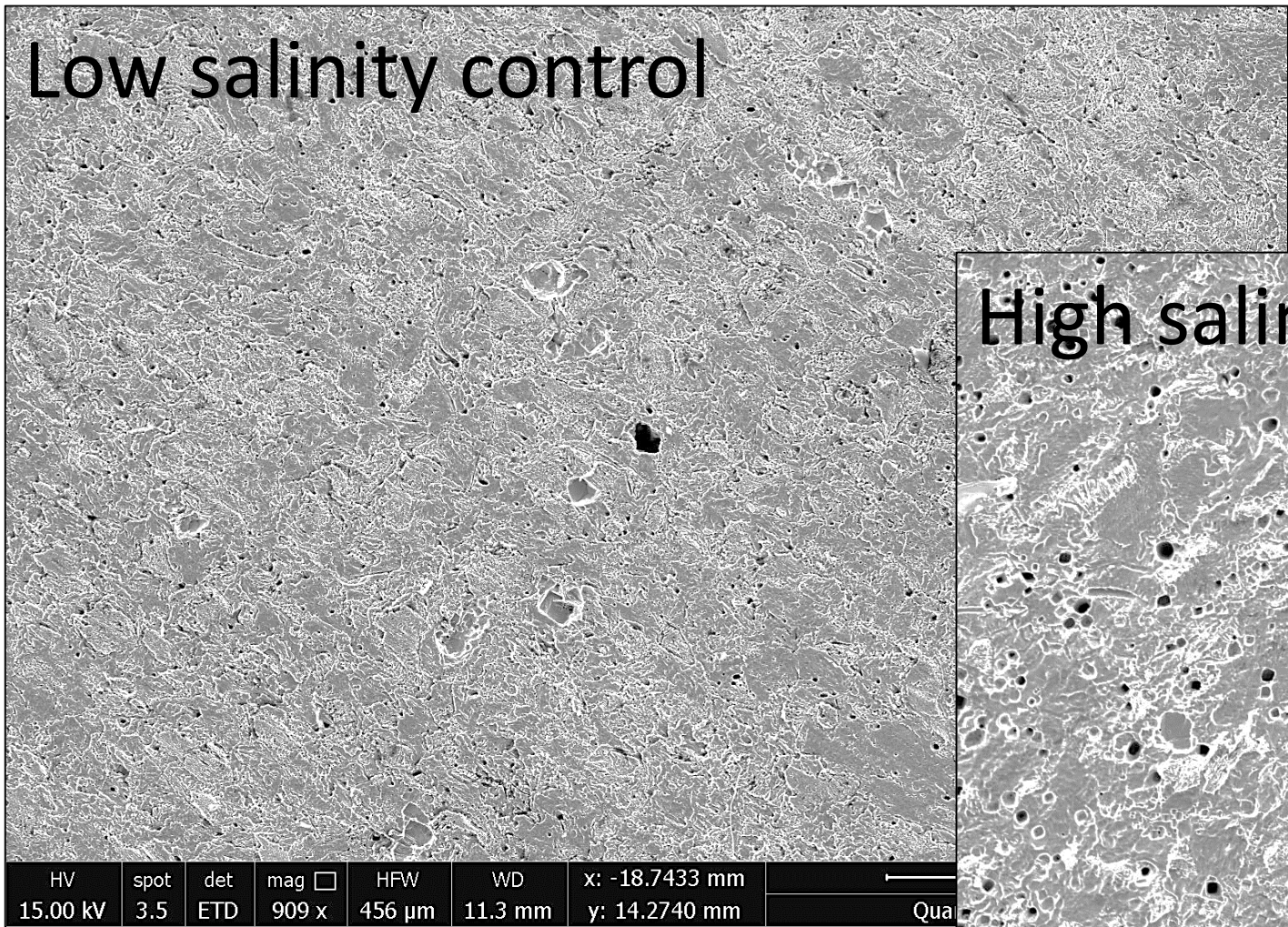
Sample	Cell numbers T0	Cell numbers T84
Bentonite	-	-
Donor amended	-	<1000
Culture amended	<1000	37,000



SRB cells/ gram sample

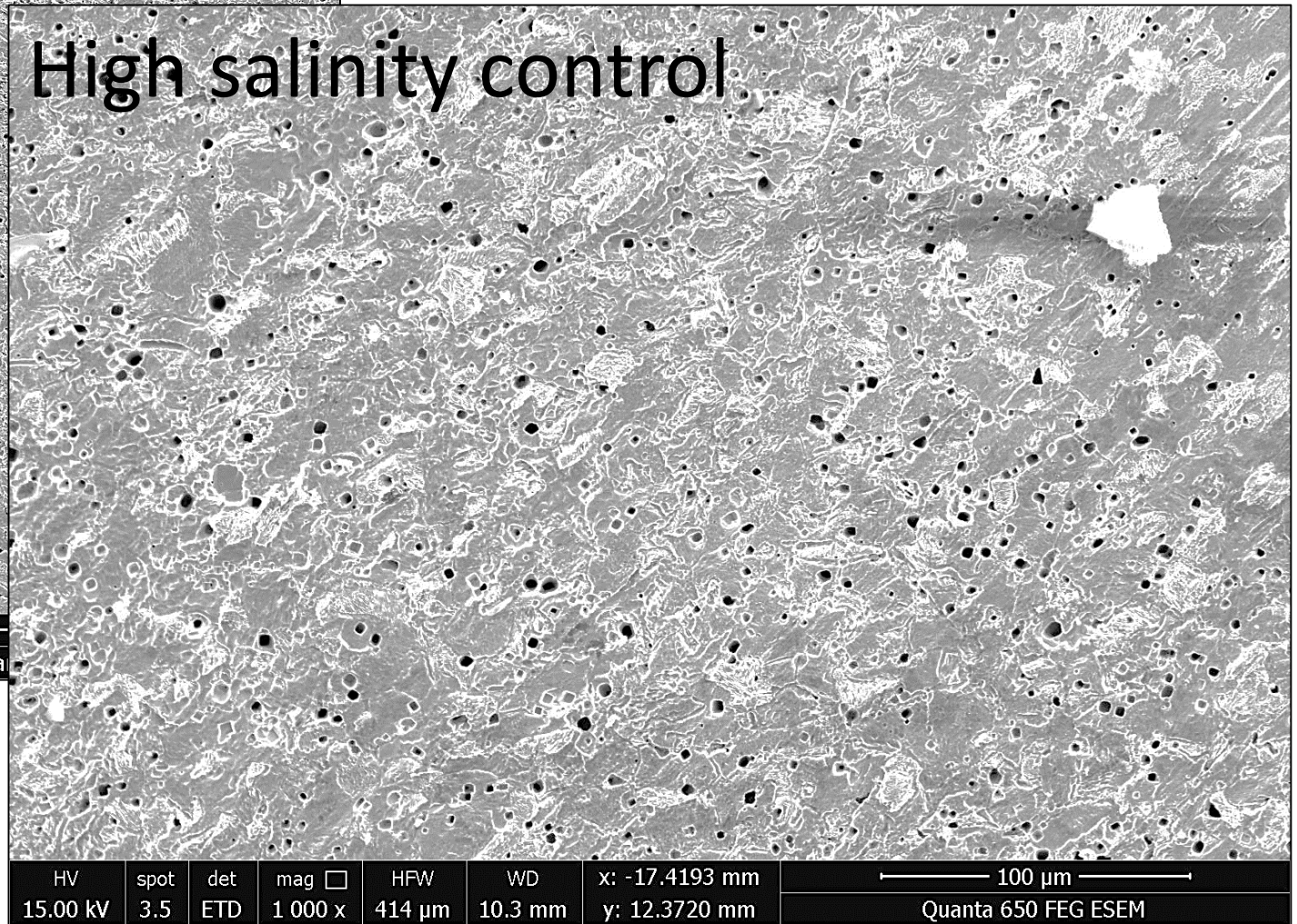
Sample	Cell numbers T0	Cell numbers T84
Bentonite	-	60,000
Donor amended	-	2,177,000
Culture amended	138,000	269,000

Low salinity control



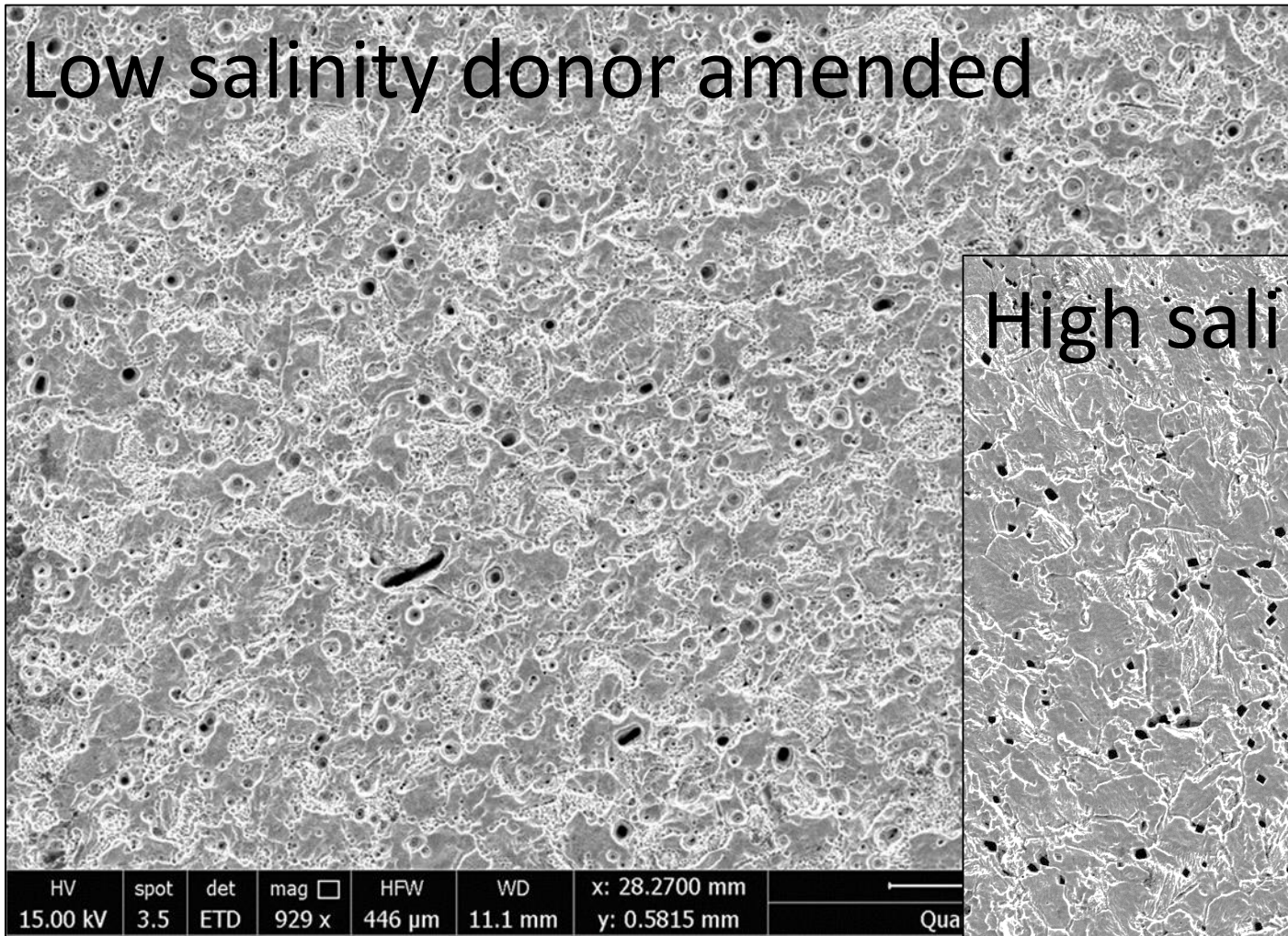
T84 Sterile steel sample at *low salinity* (washed)

High salinity control



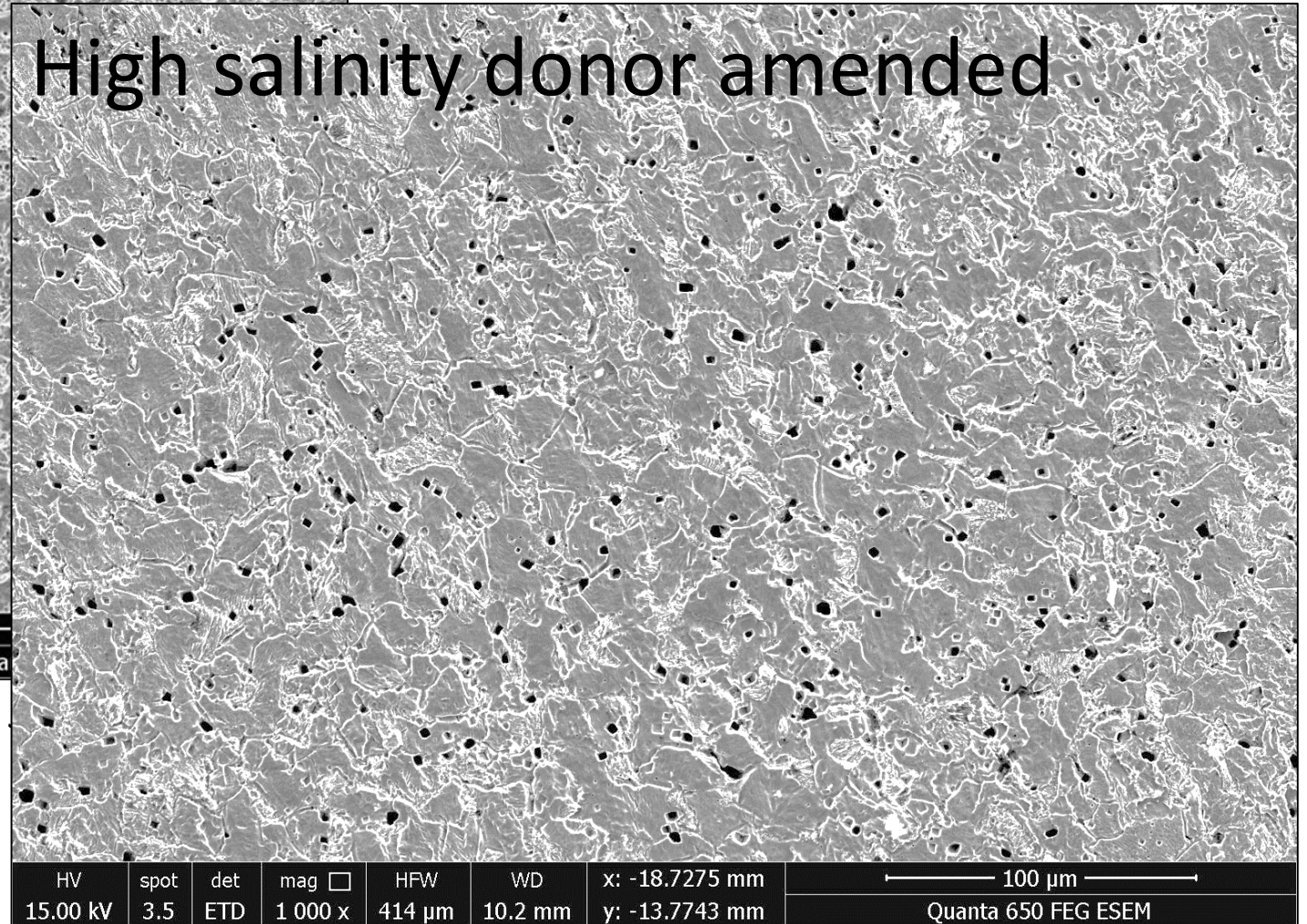
T84 Sterile microcosm steel sample at *high salinity* (washed)

Low salinity donor amended



T84 donor amended microcosm steel sample at *low*

High salinity donor amended



T84 donor amended steel sample at *high salinity* (washed)

Conclusions

- Native SRB communities extant in mined bentonites
- Proliferating bentonite SRB can enhance corrosion of steel in mobile slurries
- High salinity conditions will suppress bentonite microbial activity
- But...

Experiments do not assess the availability (and concentrations) of *in-situ electron donor* (hydrogen)



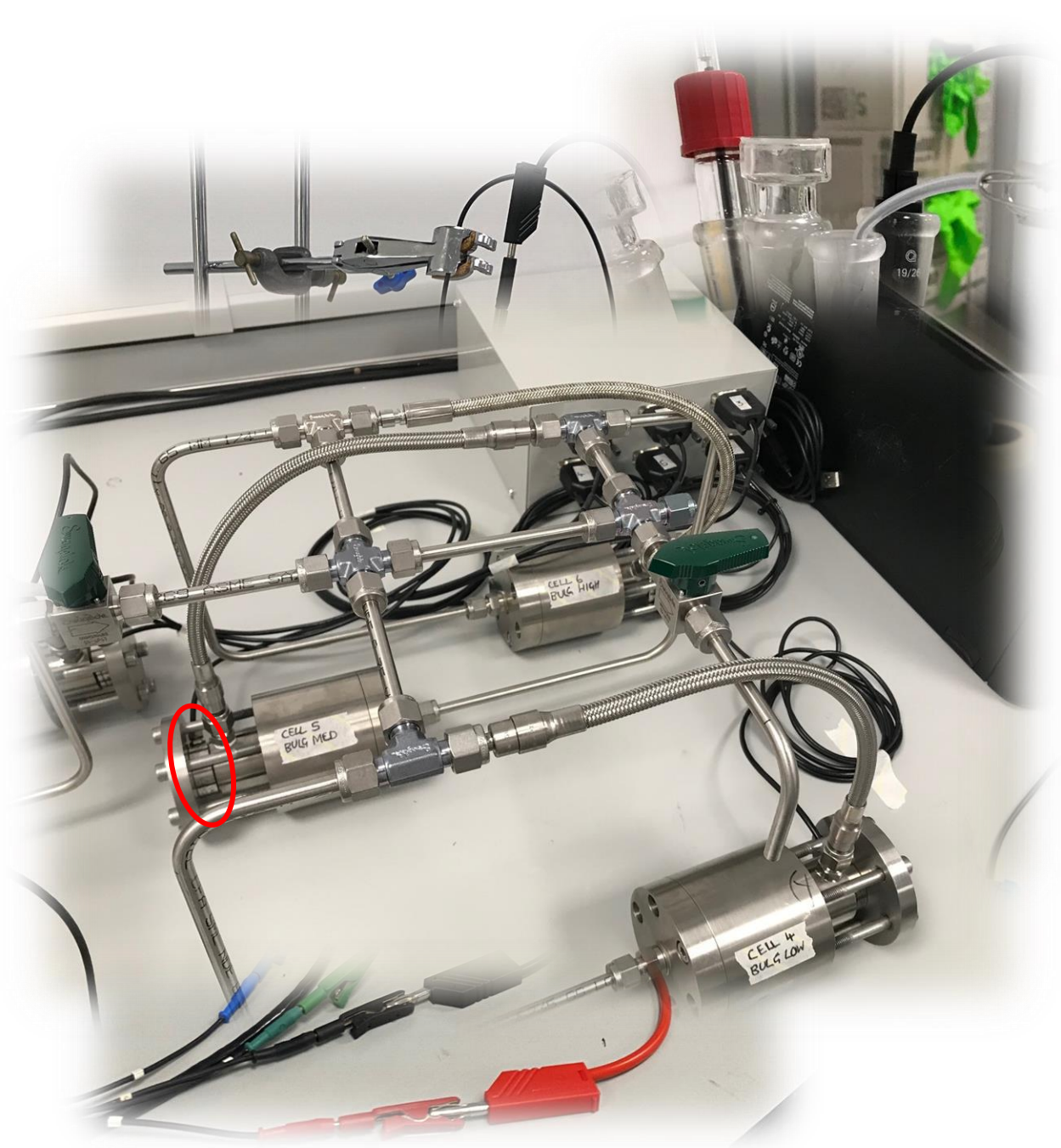
Slurry systems are not representative of a *compacted bentonite* buffer:

Rate limited transport of substrates

Lack of pore space for SRB

Future work...

- Understanding systems between different bentonites
- In-situ corrosion rates using electrochemical techniques
- Above experiments in compacted bentonite- representative systems to cover limitations from slurry experiments



Thank you for listening!

Questions?

References

- [1] Stroes-Gascoyne et al. (2010) Appl. Clay Sci. 47:155-162
- [2] Bagnoud et al. (2016) Nat Commun 7:12770
- [3] Masurat et al. (2010) Appl. Clay Sci. 47:58–64
- [4] Metcalfe et al. (2015) www.quintessa-online.com
- [5] He et al. (2020) Bull. Eng. Geol. Environ. 79:399-410
- [6] Oren (1999) Microbiol. Mol. Biol. Rev. 63(2):334-348
- [7] Haynes et al. (2021) Chem. Geol. 581:120353